

Chemistry Multiple Choice: Reactions

1. What is the general form of single displacement reactions?

- A) $A + B \rightarrow AB$ C) $AB + C \rightarrow CB + A$
B) $C_xH_y + O_2 \rightarrow CO_2 + H_2O$ D) $AB + CD \rightarrow AD + CB$

ANS: C

1. In a single displacement reaction, aqueous iron(III) sulphate reacts with barium powder. What is the balanced chemical equation for this reaction?

- A) $Fe_2(SO_4)_3(aq) + Ba(s) \rightarrow BaSO_4(s) + Fe(s)$
B) $Fe_2(SO_4)_3(aq) + 3Ba(s) \rightarrow 3BaSO_4(s) + 2Fe(s)$
C) $Fe_2(SO_4)_3(aq) + 2Ba(s) \rightarrow 2BaSO_4(s) + 3Fe(s)$
D) $FeSO_4(aq) + Ba(s) \rightarrow BaSO_4(s) + Fe(s)$

ANS: B

1. What would be the general chemical equation for decomposition reactions?

- A) $C \rightarrow A + B$ C) $AB \rightarrow A + B$
B) $AB + C \rightarrow ABC$ D) $A + O \rightarrow AOH$

ANS: C

1. What type of reaction almost always gives off large amounts of heat and light?

- A) combustion B) neutralization C) decomposition D) synthesis

ANS: A

1. What are the reactants of every hydrocarbon combustion reaction?

- A) $O_2 + H_2O$ B) $C_xH_y + O_2$ C) $C_xH_yO_x + O_2$ D) $CO_2 + H_2O$

ANS: B

1. The following balanced equation represents the reaction of aluminum with copper(II) chloride.



This is an example of a:

- A) formation reaction B) combustion reaction
C) single displacement reaction D) double displacement reaction

ANS: C

1. The products of the reaction of aqueous sodium iodide, $NaI_{(aq)}$, and aqueous lead(II) nitrate, $Pb(NO_3)_{2(aq)}$, are:

- A) $NaPb_{(s)}$ and $I_2NO_{3(aq)}$ B) $Na_{(s)}$ and $I_2NO_{3(aq)}$
C) $NaNO_{3(aq)}$ and $PbI_{2(s)}$ D) $NaNO_3(aq)$ and $PbI_{2(s)}$

ANS: D

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IIT JEE Chemistry Exam Leaders Expert, **Chapter-wise Topical Objective Study Package for CBSE 2022 Class 12 Term I Chemistry** Disha Experts, 2021-09-01. Disha's Chapter wise Topical Study Package for CBSE 2022 Class 12 Term I Chemistry is designed on the exact lines of the latest syllabus and paper pattern prescribed by the CBSE board. Circular dated July 22, 2021, for the Term I Exam to be held in November. The Book consists of a total of 6 Chapters of Term I. Each chapter is divided into 3-4 Topics. Each Topic covers exhaustive theory with Illustrations followed by an Objective Exercise consisting of MCQs, AR, Case based, VSA, SA Questions. Further, the Chapter covers Concept Maps, Important Formulae, NCERT Exemplar, Past Year Questions. The Past Solved Objective Questions covered in the book are from 2021 CBSE Sample Paper, 2020, 2019, 2018, 2017. In the end of the Chapter, an Objective Practice Exercise and a Chapter Test is provided for final practice and assessment. There are a total of 1000 Objective Questions with Solutions. The book is a One Stop Solution for Learning, Practice, Revision. *2024-25 CBSE/NIOS/ISC/UP Board 12th Class Chemistry Chapter-wise Unsolved Papers* YCT Expert Team, 2024-25 CBSE NIOS ISC UP Board 12th Class Chemistry Chapter wise Unsolved Papers 464 895 E. This book contains the previous year paper from 2010 to 2024.

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ingrédients sont le bouillon généralement préparé la veille le vin blanc l

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