

Acids and Bases

Question 1: Choose the correct answer for each of the following questions:

- 1.1 The element present in all acids...
- | | |
|------------|------------------|
| A oxygen | B carbon dioxide |
| C nitrogen | D hydrogen |
- 1.2 A sample of an acid in a beaker has a pH of 4. An alkali is added to it. What is the pH of the liquid in the beaker?
- | | |
|-------------|------------------|
| A decreases | B stays the same |
| C increases | D neutral |
- 1.3 Choose the incorrect combination:
- | | |
|-------------------------------------|---|
| A HNO_3 – Nitric acid. | B H_2SO_4 – sulphuric acid. |
| C HCl – Hydrochloric acid. | D H_2CO_3 – carbonic acid. |
- 1.4 The colour of blue litmus paper in an acid is ...
- | | |
|----------|---------|
| A yellow | B green |
| C blue | D red |
- 1.5 Which substance is always produced in the reaction between hydrochloric acid and lithium?
- | | |
|--------------|----------------------|
| A water | B hydrogen gas |
| C oxygen gas | D carbon dioxide gas |
- 1.6 Which substance is always produced in the reaction between hydrochloric acid and sodium hydroxide?
- | | |
|--------------|----------------|
| A water | B hydrogen gas |
| C oxygen gas | D precipitate |
- 1.7 Which formula represents an acid?
- | | |
|--------------------------|----------------------------|
| A KOH | B KCl |
| C CH_3OH | D CH_3COOH |
- 1.8 The method of adding an acid to a base is:
- | | |
|------------------|-----------------|
| A neutralisation | B precipitation |
| C evaporation | D decomposition |

[8]

Question 2: Write the general reaction for an acid with the following compounds:

- 2.1 Base: _____
- 2.2 Carbonate: _____
- 2.3 Metal: _____

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M. Rajshekhar



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Table of Contents Annexure Grade 9 Paper Natural Sciences 2014

1. Understanding the eBook Annexure Grade 9 Paper Natural Sciences 2014
 - The Rise of Digital Reading Annexure Grade 9 Paper Natural Sciences 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Annexure Grade 9 Paper Natural Sciences 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Annexure Grade 9 Paper Natural Sciences 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Annexure Grade 9 Paper Natural Sciences 2014
 - Personalized Recommendations
 - Annexure Grade 9 Paper Natural Sciences 2014 User Reviews and Ratings
 - Annexure Grade 9 Paper Natural Sciences 2014 and Bestseller Lists
5. Accessing Annexure Grade 9 Paper Natural Sciences 2014 Free and Paid eBooks
 - Annexure Grade 9 Paper Natural Sciences 2014 Public Domain eBooks
 - Annexure Grade 9 Paper Natural Sciences 2014 eBook Subscription Services
 - Annexure Grade 9 Paper Natural Sciences 2014 Budget-Friendly Options

6. Navigating Annexure Grade 9 Paper Natural Sciences 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Annexure Grade 9 Paper Natural Sciences 2014 Compatibility with Devices
 - Annexure Grade 9 Paper Natural Sciences 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Annexure Grade 9 Paper Natural Sciences 2014
 - Highlighting and Note-Taking Annexure Grade 9 Paper Natural Sciences 2014
 - Interactive Elements Annexure Grade 9 Paper Natural Sciences 2014
8. Staying Engaged with Annexure Grade 9 Paper Natural Sciences 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Annexure Grade 9 Paper Natural Sciences 2014
9. Balancing eBooks and Physical Books Annexure Grade 9 Paper Natural Sciences 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Annexure Grade 9 Paper Natural Sciences 2014
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Annexure Grade 9 Paper Natural Sciences 2014
 - Setting Reading Goals Annexure Grade 9 Paper Natural Sciences 2014
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Annexure Grade 9 Paper Natural Sciences 2014
 - Fact-Checking eBook Content of Annexure Grade 9 Paper Natural Sciences 2014
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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