Buoyancy Worksheet Answers

 A medical ship has a mass of 100,000 kg. What volume of fresh water (p~1025 kg/m3) will the ship displace?

100,000 [hg] / 1,025 [hg/m] - 97,56 m

2) How much mass can a 1,000,000 L bulloon lift if the inside temperature of the bulloon is 80 °C and the outside air temperature is 20 °C?

First, move the terms containing V to the same side by subtracting Vplinside Fluid from both sides:

M K V (Common Print - W (Streets Print)

Nont, plug in the numbers:

M < (1,000,000 to (1,004 a/L - 1,000 a/L)

Not be 2004 for

3) How many 10 L helium ballooms would it take to lift a man in an armeluair (75 kg) if the density of air is 1.2 g/L and the density of helium is 0.1786 g/L. Assume each balloon has a mass of 3 g. (Hint: First calculate how much extra mass a helium halloon can carry)

First like the example to the lesson, one bulbons can life 10.2 g. If each bulleon has a wass of 3 g. then each bulbons can carry 7.2 g.

TS bg / T.2 g = 18,417 Bulletons.

Answer Key Buoyancy

Joacim Rocklöv



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