

100

27. It is necessary to know the formula of a compound in order to calculate its percentage composition.

101

28. If the percent by mass of carbon in methane, CH_4 , is 75%, then 100 grams of methane contains 75 g of carbon and 25 g of hydrogen.

102

29. The formula for methane, CH_4 , is both a molecular and an empirical formula.

103

30. The empirical formula of glucose, $\text{C}_6\text{H}_{12}\text{O}_6$, is CH_2O .

C. Problems

Solve the following problems on the space provided. Show your work.

31. How many moles of Fe is 5.6 g. of 10^{23} atoms of Fe?

32. How many representative particles are in 2.33 g. 10^{23} and MgCl_2 ?

33. What is the gram molecular mass of carbon, C_7H_8 ?

34. Find the mass of each of the following:

a. 3.45 g. of 10^{23} and H_2O_2

b. 2.43 g. and H_2O_2

c. 4.50 g. and 1.30 g.

d. 25.78 g. and 8.2

Answer Key Chemical Quantities

LM Reder



Answer Key Chemical Quantities:

Critical Loads for Sulphur and Nitrogen Jan Nilsson, 1988-07-01 Milj rapport 1988 15 *Chemistry for the Biosciences* Jonathan Crowe, Tony Bradshaw, 2021 Chemistry enables our eyes to detect the world around us it determines whether something tastes sweet or sour it helps genetic information pass accurately from one generation to the next Ultimately chemistry powers life itself We don't need to dig very deep to answer the question why do biologists need chemistry Building on the success of the first three editions Chemistry for the Biosciences introduces students to all the chemistry they need to understand the biological world Renowned for its clear and straightforward explanations the book uses everyday examples and analogies throughout to help students get to grips with chemical concepts and presents them in context of biological systems wherever possible so they can see how chemistry relates to their wider studies With topics drawn from organic physical and inorganic chemistry students will encounter a broad range of essential concepts Chemistry for the Biosciences includes many learning features both in print and online to help students grasp these concepts as quickly and thoroughly as possible From the self check questions throughout each chapter to help consolidate learning to the Chemical Toolkits and Maths Tools that help students explore terminology methods and numerical skills that may be unfamiliar the book is written to be a true course companion for students on biological and biomedical science degrees one that will help them not only remember the essentials but really understand them setting students up for success in their later studies **Frank**

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