two hydrogen atoms share heir valence dectrons, both atoms achieve the electronic configuration of noble gas & statisfy the duplet rule.



Types of Covalent Bond:

Depending upon the number of bond pair, a covalent bond is further classified into three types.

- Single Covalent Bond
   Double Covalent Bond
- . Triple Covalent Bond

Single Covalent Bond (-)

A covalent bond which is formed by the mutal sharing of one bond pair is called a single covalent bond and it is represented by a single short straight line. The formation of H-H, H-CI, CH, are four examples of this type of bonding. txamples:

Formation of Chlorine Malecule:

One more electronito achieve a stable actet electronic configuration. When two chloring atoms share their valence

# **Answer Key For Convelent Bond Caoncept Map**

**McGraw-Hill Staff** 

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