

- a. Write the π -system components from the following list:
 1b, a, f, g, h
- b. Determine the ground electronic configuration: $1s^2 2s^2 2p^4$
- c. Assign the following ligand names and indicate the type of complex formed:

i. pyridine	coordination complex
ii. acetate	complexed metal-organic complex
iii. phosphate	coordination complex
iv. phosphate	complexed metal-organic complex
v. phosphate	complexed metal-organic complex
vi. phosphate	complexed metal-organic complex
- d. Name the following ligands:

i. 1,2-diaminopropane
ii. 1,3-diaminopropane
iii. 1,4-diaminopropane
iv. pyridine
v. 1,2,3,4-tetraaminoethane
vi. 1,2,3,4-tetraaminoethane
vii. 1,2,3,4-tetraaminoethane
viii. 1,2,3,4-tetraaminoethane
ix. 1,2,3,4-tetraaminoethane
x. 1,2,3,4-tetraaminoethane
xi. 1,2,3,4-tetraaminoethane
xii. 1,2,3,4-tetraaminoethane
xiii. 1,2,3,4-tetraaminoethane
xiv. 1,2,3,4-tetraaminoethane
xv. 1,2,3,4-tetraaminoethane
xvi. 1,2,3,4-tetraaminoethane
xvii. 1,2,3,4-tetraaminoethane
xviii. 1,2,3,4-tetraaminoethane
xix. 1,2,3,4-tetraaminoethane
xx. 1,2,3,4-tetraaminoethane

10. Draw condensed structural diagrams of the ligands for each of the following metals:

- i. 1,2-diaminopropane



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