

Question 1: Water at 20°C falls from a height of 854 m. If the whole energy is used in increasing the temperature, find out the final temperature.
Specific heat of water is 4200 J/K kg .

Given that

$$T_1 = 20^{\circ}\text{C} = 20 + 273 = 293 \text{ K}$$

$$\text{height} = h = 854 \text{ m}$$

$$\text{Specific heat of water} = 4200 \text{ J K}^{-1} \text{ kg}^{-1}$$

$$\text{Final temperature} = T_2 = ?$$

$$\text{P.E} = mgh \quad \text{--- (1)}$$

Solution:

We know that

$$\Delta Q = cm \Delta T \quad \text{--- (2)}$$

Whole energy is used in increasing the temperature (Given)

$$\Delta Q = \text{P.E}$$

From (1) and (2)

$$cm \Delta T = mgh$$

$$c \Delta T = gh$$

$$c (T_2 - T_1) = gh$$

$$T_2 - T_1 = \frac{gh}{c}$$

$$T_2 = \frac{gh}{c} + T_1$$

$$= \frac{9.8 \times 854}{4200} + 293$$

$$= \frac{8369.2}{4200} + 293 = 1.99 + 293$$

$$T_2 = 294.99 \text{ K} = 295 - 273^{\circ}\text{C}$$

$$T_2 = 22^{\circ}\text{C}$$

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Theoretical Chemistry from the Standpoint of Avogadro's Rule & Thermodynamics Walther Nernst, 1895 **Ice and Refrigeration**, 1892 **Gravity and the Quantum** Jasjeet Singh Bagla, Sunu Engineer, 2017-02-23 This book provides a compilation of in depth articles and reviews on key topics within gravitation cosmology and related issues It is a celebratory volume dedicated to Prof Thanu Padmanabhan Paddy the renowned relativist and cosmologist from IUCAA India on the occasion of his 60th birthday The authors many of them leaders of their fields are all colleagues collaborators and former students of Paddy who have worked with him over a research career spanning more than four decades Paddy is a scientist of

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Nieuwenhuizen,2007-10-29 In the past thirty years the area of spin glasses has experienced rapid growth including the development of solvable models for glassy systems Yet these developments have only been recorded in the original research papers rather than in a single source Thermodynamics of the Glassy State presents a comprehensive account of the modern theory of **The Mathematical Nature of the Living World** Gilbert Chauvet,2005 The topic of the book a theory of functional biology that incorporates the fundamental principles underlying the functioning of living organisms is clearly appropriate as we celebrate the 50th anniversary of the discovery by Watson and Crick of the structure of the DNA molecule

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Biothermodynamics Part A Michael L. Johnson, Jo M. Holt, Gary K. Ackers, 2009-03-14 In the past several years there has been an explosion in the ability of biologists molecular biologists and biochemists to collect vast amounts of data on their systems This volume presents sophisticated methods for estimating the thermodynamic parameters of specific protein protein DNA and small molecule interactions The use of thermodynamics in biological research is used as an energy book keeping system While the structure and function of a molecule is important it is equally important to know what drives the energy force These methods look to answer What are the sources of energy that drive the function Which of the pathways are of biological significance As the base of macromolecular structures continues to expand through powerful techniques of molecular biology such as X ray crystal data and spectroscopy methods the importance of tested and reliable methods for answering these questions will continue to expand as well

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Kinetic Theory of Nonequilibrium Ensembles, Irreversible Thermodynamics,

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Analytical Thermodynamics Dongqing Li, 2024-12-07 This second edition presents an enriched and expanded exploration of the fundamental principles of thermodynamics tailored for graduate level studies Drawing on over three decades of academic teaching experience the author has refined the content making it more accessible and comprehensive Chapter 1 has been restructured for clarity delineating Legendre Transformation and Thermodynamic Potentials into separate sections while the treatment of Chemical Potentials has been significantly augmented encompassing two component ideal gas mixtures and a re derivation of chemical potentials for dilute solutions Additionally the section on thermodynamic stability now boasts enhanced explanations and illustrative figures Chapter 2 introduces a groundbreaking section Electrolyte

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