Identifying the 5 Types of Chemical Reactions

Identify each of the reactions below as one of the 5 types of Reactions (Rxn):

- Synthesis Reaction
- Decomposition Reaction
- Single Replacement Reaction
- Double Replacement Reaction
- Combustion Reaction

Reactions (identify pattern below)

Name of Reaction Type

- 1. $2 \underbrace{Na}_{B} + \underbrace{Cl_{2}}_{A} \rightarrow 2 \underbrace{NaCl}_{A}_{B}$ (A & B combine) Ex: synthesis
- Pb + FeSO₄ → PbSO₄ + Fe
- P₄ + 3 O₂ → 2 P₂O₃
- 2 NO₂ → 2 O₂ + N₂
- Na₃PO₄ + 3 KOH → 3 NaOH + K₃PO₄
- C₃H₆O + 4 O₂ → 3 CO₂ + 3 H₂O
- MgCl₂ + Li₂CO₃ → MgCO₃ + 2 LiCl
- C₆H₁₂ + 9 O₂ → 6 CO₂ + 6 H₂O
- CaCO₃ → CaO + CO₂
- 2 AgNO₃ + Cu → Cu(NO₃)₂ + 2 Ag
- 2H₂ + O₂ → 2H₂O

Answers For Chemical Reactions

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Answers For Chemical Reactions:

Chemical Reactions, Materials and Particles Gill Murphy, 2006-03 Specials are a teaching resource designed for KS3 students whose literacy skills are considerably lower than their age These books have an older format to counteract the simple text and cover various topics They include activities visuals and assessment sheets as well as teacher pages

Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day

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