

A chemical bond - forms when 2 or more atoms rearrange _____
_____ to increase _____.

ionic bond - forms when valence _____ and _____
from one atom to another.

cation - atom _____ electrons to become _____ charged.

anion - atom _____ electrons to become _____ charged.

In ionic compounds the ions are arranged in a _____
_____ forces hold the ions together.

Properties of ionic compounds:

- High _____ and _____ points
- _____ - not easily _____
- _____ electricity when _____ or
_____ because the ions are free to _____.

covalent bond - _____ and _____, forming
_____.

Covalent compounds have _____ forces holding the
_____ together.

Properties of covalent compounds:

- lower _____ and _____ points
- Many covalent compounds are _____ liquids or gases.
- _____ - easier to _____
- are not _____ of electricity

electronegativity - property that tells how strong an atom's _____
is for _____.

Since oxygen has a _____ electronegativity than hydrogen, oxygen
holds onto shared electrons _____, giving the oxygen a _____
negative charge and the hydrogen a partial _____ charge.

Answers For Note Taking Guide Episode 501

Michael Seilmaier



Answers For Note Taking Guide Episode 501:

Answers For Note Taking Guide Episode 501 Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Answers For Note Taking Guide Episode 501**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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