## 0.1: Physical Constants

$c   3 \times 10^8 \text{ m/s}$	c	Speed of light
$h = 6.63 \times 10^{-34} \text{ J s}$	h	Planck constant
hc 1242 eV-nm	hc	
$G = 6.67 \times 10^{-11} \text{ m}^3 \text{ kg}^{-1} \text{ s}^{-2}$	G	Gravitation constant
$k = 1.38 \times 10^{-23} \text{ J/K}$	$\boldsymbol{k}$	Boltzmann constant
R 8.314 J/(mol K)	$\boldsymbol{R}$	Molar gas constant
$N_{\rm A}$ 6.023 × 10 <sup>23</sup> mol <sup>-1</sup>	$N_{A}$	Avogadro's number
e 1.602 × 10 <sup>-19</sup> C	e	Charge of electron
$-\mu_0$ $4\pi \times 10^{-7} \text{ N/A}^2$	$\mu_0$	Permeability of vac-
WAS COMMENTED TO THE PROPERTY OF THE PROPERTY		uum
$\epsilon_0 = 8.85 \times 10^{-12} \text{ F/m}$	$\epsilon_0$	Permitivity of vacuum
$\frac{1}{4760}$ 9 × 10 <sup>9</sup> N m <sup>2</sup> /C <sup>2</sup>		Coulomb constant
F 96485 C/mol		Faraday constant
$m_e$ 9.1 × 10 <sup>-31</sup> kg	$m_c$	Mass of electron
$m_p = 1.6726 \times 10^{-27} \text{ kg}$	$m_p$	Mass of proton
$m_n = 1.6749 \times 10^{-27} \text{ kg}$	$m_n$	Mass of neutron
$u = 1.66 \times 10^{-27} \text{ kg}$	u	Atomic mass unit
$u = 931.49 \text{ MeV/c}^2$	u	Atomic mass unit
$\sigma = 5.67 \times 10^{-8} \text{ W/(m}^2 \text{ K}^4)$	$\sigma$	Stefan-Boltzmann
		constant
$R_{\infty}$ 1.097 × 10 <sup>7</sup> m <sup>-1</sup>	$R_{\infty}$	Rydberg constant
$\mu_B = 9.27 \times 10^{-24} \text{ J/T}$	$\mu_B$	Bohr magneton
$a_0 = 0.529 \times 10^{-10} \text{ m}$	$a_0$	Bohr radius
atm $1.01325 \times 10^5 \text{ Pa}$	atm	Standard atmosphere
t . b $2.9 \times 10^{-3} \text{ m K}$	b	Wien displacement
OIFOI LITEOUV		
$F$ 96485 C/mol $m_e$ 9.1 × 10 <sup>-31</sup> kg $m_p$ 1.6726 × 10 <sup>-27</sup> k $m_n$ 1.6749 × 10 <sup>-27</sup> kg $m_n$ 1.66 × 10 <sup>-27</sup> kg $m_n$ 931.49 MeV/c <sup>2</sup> $\sigma$ 5.67 × 10 <sup>-8</sup> W/(m <sup>2</sup> $R_{\infty}$ 1.097 × 10 <sup>7</sup> m <sup>-1</sup> $m_n$ 9.27 × 10 <sup>-24</sup> J/ $m_n$ 0.529 × 10 <sup>-10</sup> m atm 1.01325 × 10 <sup>5</sup> Pa	$F$ $m_e$ $m_p$ $m_n$ $u$ $\sigma$ $R_{\infty}$ $\mu_B$ $a_0$	Faraday constant Mass of electron Mass of proton Mass of neutron Atomic mass unit Atomic mass unit Stefan-Boltzmann constant Rydberg constant Bohr magneton Bohr radius Standard atmosphere

# All Constant That Are Used In Building Science N

**R Sandford** 

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NBS Building Science Series United States. National Bureau of Standards,1978 NBS Building Science Series

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