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# TYPES OF CHEMICAL REACTIONS

identify the type of chemical reaction (synthesis, decomposition, single displacement or replacement, double displacement or replacement).

1	$AB + C \rightarrow AC + B$	Single replacement
2	AB + CD AD + CB	Double replacement
3	$AB \rightarrow A + B$	Decomposition
4	$A + B \rightarrow AB$	Synthesis
5		Decomposition
6		Double replacement
7	○ • ○ ○ • ○ • ○	Single replacement
8	→	Synthesis
9	CaCO <sub>3</sub> → CaO + CO <sub>2</sub>	Decomposition
10	NaCl + AgNO <sub>3</sub> → NaNO <sub>2</sub> + AgCl	Double replacement
11	$Zn + 2HCl \rightarrow ZnCl_2 + H_2$	Single replacement
12	$Na(s) + Cl_2(g) \rightarrow 2NaCl(s)$	Synthesis
13	HCl + NaOH → NaCl + H <sub>2</sub> O	Double replacement
14	$2H_2O \rightarrow 2H_2 + O_2$	Decomposition
15	$Zn(s) + CuSO_a(aq) \rightarrow ZnSO_a(aq) + Cu(s)$	Single replacement

# **Answers For Types Of Reactions**

William C. Groutas, Athri D. Rathnayake

## **Answers For Types Of Reactions:**

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