

Post-Lab Questions

1. From the experiment, which reactant is the limiting reactant? Which one is the reactant in excess? How do you determine which one is the limiting reactant?
2. If you weigh the watch glass then write your name on it with the wax pencil, how will that affect your percent yield?
3. What factors could cause the percent yield of a product to be less than 100%?
4. What factors could cause the percent yield of a product to be more than 100%?
5. Balance the following reaction: $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 - a. 4.30 g of C_3H_8 reacts with 4.30 g of O_2 . Which one is the limiting reagent, and which one is in excess? Explain your reasoning.
 - b. Based on the limiting reagent, what are the theoretical yields of CO_2 and H_2O ?
 - c. You perform the experiment and produce 2.67 g of CO_2 and 0.79 g of H_2O . What are your percent yields for each product?

Answers Key To Chemistry 1211 Lab Manual

Wolfgang Guggemos



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Unveiling the Magic of Words: A Report on "**Answers Key To Chemistry 1211 Lab Manual**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Answers Key To Chemistry 1211 Lab Manual**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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