



NAME: Abhishek H. Nair Div: A Roll No. 1

SCIENCE Paper – 2

(Two hours)

You will **not** be allowed to write during the first 15 minutes.

This time is to be spent in reading the Question Paper.

The time given at the head of this paper is the time allowed for writing the answers.

Section I is compulsory. Attempt any four questions from Section II.

The intended marks for questions or parts of questions are given in brackets []

SECTION I (40 Marks)

(Attempt ALL questions)

Question 1

[15]

Choose the correct answer from the options given below:

(Do not copy the question; write the correct answers only.)

- Whenever the gases react under similar conditions of temperature and pressure, they do so in volumes which bear a simple ratio of each other and the products if gaseous. The above law was stated by
 a) Boyle b) Charles c) Gay Lussac d) Avogadro
- A strip of copper is placed in four different colourless salt solutions. After electrolysis, one of the salt solutions becomes blue in colour. Identify the salt solution which turns blue from the given salt solutions:
 a) K_2SO_4 b) $CuSO_4$ c) $AgSO_4$ d) $CaSO_4$
- Potassium is the element below sodium in the long form of the periodic table. The electronegativity of potassium will be
 a) Lower than sodium b) Higher than sodium c) Same as sodium d) None of these
- The salt solution which does not react with ammonium hydroxide is :
 a) Calcium Nitrate b) Zinc Nitrate c) Lead Nitrate d) Copper Nitrate
- Assertion :** Sodium is the most reactive metal among group 1 elements.
Reason : In case of metals, greater the tendency to lose electrons, greater is the reactivity.
 a) A and R both are correct statements and R is the correct explanation of the A
 b) A and R both are correct statements but R is not the correct explanation of the A
 c) A is true, but R is false
 d) A is false , but R is true



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Answers For A Ls Chemistry

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Chemistry Questions - Answers for Intermediate Mr. Ajeet Singh, 2018-05-13 A question answer formatted book designed for intermediate level chemistry students covering theory based and numerical questions to aid learning and exam preparation

New Pattern NTA JEE Main Quick Guide in Chemistry with Numeric Answer Questions 3rd Edition Disha Experts, 2019-10-01 As NTA introduces Numeric Answer Questions in JEE Main Disha launches the Questions the 3rd latest updated edition of New Pattern NTA JEE Main Quick Guide in Chemistry with Numeric Answer Questions This study material is developed for quick revision and practice of the complete syllabus of the JEE Main Exam in a short span of 40 days The book can prove to be the ideal material for class 12 students as they can utilise this book to revise their preparation immediately after the board exams The book contains 27 chapters of class 11 12 and each Chapter contains JEE Main 6 Years at a Glance i e JEE Main 2019 2014 with TOPIC WISE Analysis Detailed Concept Maps covers entire JEE Syllabus for speedy revision IMPORTANT CRITICAL Points of the Chapter for last minute revision TIPS to PROBLEM SOLVING to help students to solve Problems in shortest possible time Exercise 1 CONCEPT BUILDER A Collection of Important Topic wise MCQs to Build Your Concepts Exercise 2 CONCEPT APPLICATOR A Collection of Quality MCQs that helps sharpen your concept application ability Exercise 3 Numeric Answer Questions A Collection of Quality Numeric Answer Questions as per the new pattern of JEE Answer Keys Detailed Solutions of all the Exercises and Past years problems are provided at the end of the chapter

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Answer Book for Chemistry, Sixth Edition Charles E. Mortimer, 1986

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Publishers' Circular and Booksellers' Record of British and Foreign Literature, 1894 *The Athenaeum*, 1874 *Chemical News and Journal of Physical Science*, 1863 *Chemical News and Journal of Industrial Science*, 1893 *Chemistry in Daily Life* Dr. Benvikram Barman, Dr. Bhupendra Singh Banjare, Dr. Dolly Baghel, Dr. Dinesh Kumar Tandan, Mr. Naveen Verma, Mr. Rupendra Kumar Sahu, 2025-08-12 Chemistry in Daily Life is a comprehensive textbook designed to connect fundamental chemical principles with everyday applications In this book have 23 chapter each chapter explain with suitable figure and in last section of each chapter have multiple type question with answer This book explores how chemistry influences food environment health and industry It covers the composition and quality analysis of dairy products beverages food additives including artificial sweeteners Flavors preservatives colorants and contaminants Chapters paints dyes pigments and their industrial applications Air and water pollution are observed finished topics like air pollutants greenhouse gases acid rain and wastewater treatment Readers gain understanding into water purification methods the hydrologic cycle and soil chemistry The role of fertilizers and their classifications is explained alongside the chemistry of essential biomolecules like vitamins carbohydrates oils Medicinal chemistry is detailed discussions on analgesics antibiotics antimalarials other drugs The book also introduces concepts in chemical kinetics thermodynamics and energy systems including fuel cells batteries solar energy Finally environmental challenges such as plastic waste heavy metal pollution are addressed with eco friendly solutions Rich in real life examples diagrams exercises this book is perfect for students educators and anyone interested how chemistry shapes our world

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