# 1999 Advanced Placement Chemistry Exam Section I: Multiple Choice

Time — I hour and 30 minutes

#### NO CALCULATORS MAY BE USED WITH SECTION I.

Note: For all questions, assume that the temperature is 298 K, the pressure is 1.00 atmosphere, and solutions are aqueous unless otherwise specified.

Throughout the test the following symbols have the definitions specified unless otherwise noted.

7	- temperature
	- pressure
W	= volume
S	= entropy
	- enthalpy
	- free energy
	= molar gas constant
70	<ul> <li>number of moles</li> </ul>
A16	= molar

481	-	molal
L, mL		liter(s), milliliter(s)
g		gram(s)
nm	100	nanometer(s)
action	100	atmosphere(s)
J, kJ		joule(s), kilojoule(s)
W		volt(s)
med		mole(s)

#### Part A

<u>Directions</u>: Each set of lettered choices below refers to the numbered statements immediately following it. Select the one lettered choice that best fits each statement and then fill in the corresponding oval on the answer sheet. A choice may be used once, more than once, or not at all in each set.

Questions 1-4 refer to the following types of energy.

- (A) Activation energy
- (B) Free energy
- (C) Ionization energy
- (D) Kinetic energy
- (E) Lattice energy
- The energy required to convert a ground-state atom in the gas phase to a gaseous positive ion
- The energy change that occurs in the conversion of an ionic solid to widely separated gaseous ions
- The energy in a chemical or physical change that is available to do useful work
- The energy required to form the transition state in a chemical reaction

Questions 5-8 refer to atoms for which the occupied atomic orbitals are shown below.

- (A) 1s \_ 2s \_
- (B) 1s \_\_ 2s \_\_
- (C) 1s \_ 2s \_ 2p \_ \_ \_
- (D) 1s \_ 2s \_ 2p \_ \_ \_ \_
- (E) [Ar] 4s \_ 3d \_ \_ \_ \_ \_ \_
- 5. Represents an atom that is chemically unreactive
- Represents an atom in an excited state
- 7. Represents an atom that has four valence electrons
- 8. Represents an atom of a transition metal

Questions 9-12 refer to aqueous solutions containing 1:1 mole ratios of the following pairs of substances. Assume all concentrations are 1 M.

# **2008 Ap Chemistry Exam Multiple Choice Answers**

**M** Carnoy

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#### **Table of Contents 2008 Ap Chemistry Exam Multiple Choice Answers**

- 1. Understanding the eBook 2008 Ap Chemistry Exam Multiple Choice Answers
  - The Rise of Digital Reading 2008 Ap Chemistry Exam Multiple Choice Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying 2008 Ap Chemistry Exam Multiple Choice Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - $\circ \ \ Determining \ Your \ Reading \ Goals$
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 2008 Ap Chemistry Exam Multiple Choice Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from 2008 Ap Chemistry Exam Multiple Choice Answers
  - Personalized Recommendations
  - 2008 Ap Chemistry Exam Multiple Choice Answers User Reviews and Ratings

- 2008 Ap Chemistry Exam Multiple Choice Answers and Bestseller Lists
- 5. Accessing 2008 Ap Chemistry Exam Multiple Choice Answers Free and Paid eBooks
  - 2008 Ap Chemistry Exam Multiple Choice Answers Public Domain eBooks
  - 2008 Ap Chemistry Exam Multiple Choice Answers eBook Subscription Services
  - 2008 Ap Chemistry Exam Multiple Choice Answers Budget-Friendly Options
- 6. Navigating 2008 Ap Chemistry Exam Multiple Choice Answers eBook Formats
  - o ePub, PDF, MOBI, and More
  - 2008 Ap Chemistry Exam Multiple Choice Answers Compatibility with Devices
  - 2008 Ap Chemistry Exam Multiple Choice Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 2008 Ap Chemistry Exam Multiple Choice Answers
  - Highlighting and Note-Taking 2008 Ap Chemistry Exam Multiple Choice Answers
  - Interactive Elements 2008 Ap Chemistry Exam Multiple Choice Answers
- 8. Staying Engaged with 2008 Ap Chemistry Exam Multiple Choice Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 2008 Ap Chemistry Exam Multiple Choice Answers
- 9. Balancing eBooks and Physical Books 2008 Ap Chemistry Exam Multiple Choice Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 2008 Ap Chemistry Exam Multiple Choice Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine 2008 Ap Chemistry Exam Multiple Choice Answers
  - Setting Reading Goals 2008 Ap Chemistry Exam Multiple Choice Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 2008 Ap Chemistry Exam Multiple Choice Answers
  - Fact-Checking eBook Content of 2008 Ap Chemistry Exam Multiple Choice Answers
  - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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