

2008 AP MC Test

Section I
Part A

CHEMISTRY

Section I

Time—1 hour and 30 minutes

NO CALCULATOR MAY BE USED WITH SECTION I.

Note: For all questions, assume that the temperature is 298 K, the pressure is 1.00 atmosphere, and solutions are aqueous unless otherwise specified.

Throughout the test the following symbols have the definitions specified unless otherwise noted.

| | | | |
|-----|----------------------|-------|---------------------------|
| T | = temperature | L, mL | = liter(s), milliliter(s) |
| P | = pressure | g | = gram(s) |
| V | = volume | nm | = nanometer(s) |
| S | = entropy | atm | = atmosphere(s) |
| H | = enthalpy | mm Hg | = millimeters of mercury |
| G | = Gibbs free energy | J, kJ | = joule(s), kilojoule(s) |
| R | = molar gas constant | V | = volt(s) |
| n | = number of moles | mol | = mole(s) |
| M | = molar | | |
| m | = molal | | |

Part A

Directions: Each set of lettered choices below refers to the numbered statements immediately following it. Select the one lettered choice that best fits each statement and then fill in the corresponding oval on the answer sheet. A choice may be used once, more than once, or not at all in each set.

Questions 1-3 refer to the following types of elements in the periodic table.

- (A) Noble gases
- (B) Alkali metals
- (C) Halogens
- (D) Transition elements
- (E) Actinides

1. Are the most difficult to oxidize in a given period of the periodic table
2. Are always radioactive
3. Are the most likely to form anions

2008 Ap Chemistry Multiple Choice Answer Key

L Darling-Hammond



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