#### Unit 08 - Stoichiometry - Worksheet 1

Answer the following questions in the space provided at the left.

18 mol O<sub>2</sub> 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?

2KCLO<sub>3</sub> → 2KCl + 3O<sub>2</sub>

12 mol KClO<sub>3</sub> 3 mol O<sub>2</sub> 2 mol KClO<sub>3</sub>

21 mol O<sub>2</sub> 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?

2KCLO<sub>3</sub> → 2KCl + 3O<sub>2</sub>

14 mol KetO<sub>3</sub> 3 mol O<sub>2</sub> 2 mol KClO<sub>3</sub>

\_\_45 mol H<sub>2</sub>O\_\_\_ 3. How many moles of water are produced from 22.5 moles of oxygen?
2H<sub>2</sub> + O<sub>1</sub>Z → 2H<sub>2</sub>O

22.5 mol-O<sub>2</sub> 2 mol H<sub>2</sub>O 1 mol-O<sub>2</sub>

\_20 mol HCI\_\_\_\_ 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.

H<sub>2</sub> + Cl<sub>2</sub> → 2HCl

10 mol H<sub>2</sub> 2 mol HCl 1 mol H<sub>2</sub>

21 mol Cl<sub>2</sub> 5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.

2Fe + 3Cl<sub>2</sub> → 2FeCl<sub>3</sub>

14 mol EeCl<sub>3</sub> 3 mol Cl<sub>2</sub> 2 mol FeCl<sub>3</sub>

# **Answer Sheet To Chemistry Quiz Review Stoichiometry**

**Arshad Iqbal** 

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