

# Periodic Table Questions and Answers

How many elements are there?

118

What is the periodic table?

A chart that organizes elements by atomic number and shared properties

Who invented the periodic table?

Dmitri Mendeleev

A chart that organizes elements by atomic number and shared properties

|    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |     |    |     |    |     |    |     |    |    |    |    |    |    |    |
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| 1  | H  | 2  | He | 3  | Li | 4  | Be | 5  | B  | 6  | C  | 7  | N  | 8  | O  | 9  | F  | 10 | Ne |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |     |    |     |    |     |    |     |    |    |    |    |    |    |    |
| 11 | Na | 12 | Mg | 13 | Al | 14 | Si | 15 | P  | 16 | S  | 17 | Cl | 18 | Ar | 19 | K  | 20 | Ca | 21 | Sc | 22 | Ti | 23 | V  | 24 | Cr | 25 | Mn | 26 | Fe | 27 | Co | 28 | Ni | 29 | Cu | 30 | Zn | 31 | Ga | 32 | Ge | 33 | As | 34 | Se | 35 | Br | 36 | Kr |    |    |    |    |     |    |     |    |     |    |     |    |    |    |    |    |    |    |
| 37 | Rb | 38 | Sr | 39 | Y  | 40 | Zr | 41 | Nb | 42 | Mo | 43 | Tc | 44 | Ru | 45 | Rh | 46 | Pd | 47 | Ag | 48 | Cd | 49 | In | 50 | Sn | 51 | Sb | 52 | Te | 53 | I  | 54 | Xe | 55 | Cs | 56 | Ba | 57 | La | 58 | Ce | 59 | Pr | 60 | Nd | 61 | Pm | 62 | Sm | 63 | Eu | 64 | Gd | 65  | Tb | 66  | Dy | 67  | Ho | 68  | Er | 69 | Tm | 70 | Yb | 71 | Lu |
| 72 | Hf | 73 | Ta | 74 | W  | 75 | Re | 76 | Os | 77 | Ir | 78 | Pt | 79 | Au | 80 | Hg | 81 | Tl | 82 | Pb | 83 | Bi | 84 | Po | 85 | At | 86 | Rn | 87 | Fr | 88 | Ra | 89 | Ac | 90 | Th | 91 | Pa | 92 | U  | 93 | Np | 94 | Pu | 95 | Am | 96 | Cm | 97 | Bk | 98 | Cf | 99 | Es | 100 | Fm | 101 | Md | 102 | No | 103 | Lr |    |    |    |    |    |    |

periodic  
nds?  
r, ionization

Why a  
They have

What are the periodic table trends?

electronegativity, ionization energy, atomic radius, electron affinity

Why are noble gas so inert?

They have stable electron configurations.

# Answers For Exploring The Periodic Table

**AW Rasmussen**



## Answers For Exploring The Periodic Table:

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Michael F. Ashby, Hugh Shercliff, David Cebon, 2013-12-03 Materials Engineering Science Processing and Design winner of a 2014 Textbook Excellence Award Texty from The Text and Academic Authors Association is the ultimate materials engineering text and resource for students developing skills and understanding of materials properties and selection for engineering applications Written by world class authors it takes a unique design-led approach that is broader in scope than other texts thereby meeting the curriculum needs of a wide variety of courses in the materials and design field from introduction to materials science and engineering to engineering materials materials selection and processing and materials in design This new edition retains its design-led focus and strong emphasis on visual communication while expanding its treatment of crystallography and phase diagrams and transformations to fully meet the needs of instructors teaching a first year course in materials The book is fully linked with the leading materials software package used in over 600 academic institutions worldwide as well as numerous government and commercial engineering departments Winner of a 2014 Texty Award from the Text and Academic Authors Association Design-led approach motivates and engages students in the study of

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    - Chapter 2: Essential Elements of Answers For Exploring The Periodic Table
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    - Chapter 4: Answers For Exploring The Periodic Table in Specific Contexts
    - Chapter 5: Conclusion
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