



Interview Chemistry Questions And Answers Guide

- Q1:** What is the difference between a chemical reaction and a physical change?
- A1:** A chemical reaction involves the breaking and forming of chemical bonds, resulting in the formation of new substances with different chemical properties. A physical change, on the other hand, involves a change in the state or form of a substance without any change in its chemical composition.
- Q2:** What is the difference between a mixture and a compound?
- A2:** A mixture is a combination of two or more substances that are not chemically bonded together. The components of a mixture retain their individual chemical properties. A compound, on the other hand, is a substance formed by the chemical combination of two or more elements in a fixed ratio. The components of a compound lose their individual chemical properties and form a new substance with different chemical properties.
- Q3:** What is the difference between a solution and a suspension?
- A3:** A solution is a homogeneous mixture of two or more substances. The components of a solution are uniformly distributed and cannot be separated by physical means. A suspension, on the other hand, is a heterogeneous mixture of two or more substances. The components of a suspension are not uniformly distributed and can be separated by physical means.
- Q4:** What is the difference between a solid, liquid, and gas?
- A4:** A solid is a state of matter in which the particles are closely packed and have a fixed shape and volume. A liquid is a state of matter in which the particles are loosely packed and have a fixed volume but no fixed shape. A gas is a state of matter in which the particles are widely spaced and have no fixed shape or volume.
- Q5:** What is the difference between a pure substance and a mixture?
- A5:** A pure substance is a substance that has a constant chemical composition and cannot be separated into two or more substances by physical means. A mixture, on the other hand, is a substance that can be separated into two or more substances by physical means.
- Q6:** What is the difference between a chemical reaction and a physical change?
- A6:** A chemical reaction involves the breaking and forming of chemical bonds, resulting in the formation of new substances with different chemical properties. A physical change, on the other hand, involves a change in the state or form of a substance without any change in its chemical composition.
- Q7:** What is the difference between a mixture and a compound?
- A7:** A mixture is a combination of two or more substances that are not chemically bonded together. The components of a mixture retain their individual chemical properties. A compound, on the other hand, is a substance formed by the chemical combination of two or more elements in a fixed ratio. The components of a compound lose their individual chemical properties and form a new substance with different chemical properties.
- Q8:** What is the difference between a solution and a suspension?
- A8:** A solution is a homogeneous mixture of two or more substances. The components of a solution are uniformly distributed and cannot be separated by physical means. A suspension, on the other hand, is a heterogeneous mixture of two or more substances. The components of a suspension are not uniformly distributed and can be separated by physical means.
- Q9:** What is the difference between a solid, liquid, and gas?
- A9:** A solid is a state of matter in which the particles are closely packed and have a fixed shape and volume. A liquid is a state of matter in which the particles are loosely packed and have a fixed volume but no fixed shape. A gas is a state of matter in which the particles are widely spaced and have no fixed shape or volume.
- Q10:** What is the difference between a pure substance and a mixture?
- A10:** A pure substance is a substance that has a constant chemical composition and cannot be separated into two or more substances by physical means. A mixture, on the other hand, is a substance that can be separated into two or more substances by physical means.

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