- Alkanols can be manufactured from alkenes by the initial reaction of alkenes with
  - A. bromine in tetrachloromethane.
  - B. concentrated tetraoxosulphate(VI) acid.
  - c. aqueous potassium tetraoxomanganate(VII).
  - D. sodium hydroxide solution.
- 8. Which of the following statements about the standard hydrogen electrode is not correct?
  - A. The hydrogen gas is at a pressure of 1 atmosphere
  - A solution containing 1 mol dm<sup>-3</sup> of H<sup>+</sup>(mo) ions is used
  - C. A platinum electrode is used
  - D. The temperature is kept at 20 °C
- If 60 g of M combines with 24 g of oxygen, what would the empirical formula of the oxide be?
   [ O = 16.0, M = 120]
  - A. MO
  - B. MO<sub>2</sub>
  - C. MO<sub>3</sub>
  - D. MO.
- The products of the electrolysis of dilute sodium chloride using carbon electrodes are

	Amade	1	Corbode
Α	chlorine	and	sodium.
B.	oxygen	amd 1	hydrogen.
C.	chlorine	and	hydrogen.
D	hydrogen	and	oxygen.

- Determine the quantity of electricity used when a current of 0.20 amperes is passed through an electrolytic cell for 60 minutes.
  - A. 12 C
  - B. 120 C
  - C. 360 C
  - D. 720 C

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- Oxochlorate(I) acid is used as a bleaching, agent because it is
  - A. a weak acid.
  - B. a reducing agent.
  - C. an oxidizing agent.
  - D. a strong acid.
- The IUPAC name for CH<sub>3</sub>CHCH<sub>3</sub>CHCICHCH<sub>3</sub>CH<sub>3</sub>CH<sub>3</sub> is
  - A. 2, 4 dimethyl 3 chlorohexane.
  - B. 3, 5 dimethyl 4 chlorobexane.
  - C. 4 chloro 3, 5 dimethylhexane.
  - D. 3 chloro 2, 4 dimethylhexane.
- 14. A colourless gas with a pungent smell is evolved when dilute hydrochloric acid is added to a sample of a salt. The gas evolved could turn
  - A acidified K<sub>2</sub>Cr<sub>2</sub>O<sub>2</sub> solution colourless.
  - B. acidified KMnO<sub>4</sub> solution colourless.
  - C. Fe(NO<sub>1</sub>)<sub>2</sub> solution green.
  - D. Pb(NO<sub>3</sub>)<sub>3</sub> paper black.
- 15. If 5.0 g of marble reacts with 25.0 cm<sup>3</sup> hydrochloric acid, which of the following combinations has the fastest reaction rate?
  - A. Marble chips and 2.0 mol dur 3 HCl(oa)
  - B. Marble chips and 2.5 mol dnr3 HCl<sub>(eq)</sub>
  - C. Powdered marble and 2.5 mol dm<sup>-3</sup> HCl<sub>(ac)</sub>
  - D. Powdered marble and 2.0 mol dm<sup>-3</sup> HCl<sub>(an)</sub>

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**Ransford Tetteh** 

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