



OWNER'S HANDBOOK

FOR

1971 MODELS
30 CU. IN. [**500** C.C.] TWINS

DAYTONA T100R
TROPHY 500 T100 C

U.S.A. EDITION

7 Triumph Daytona Owners Manual

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be

22 414 l we can substitute 101 325 kPa 101 325 kPa for pressure 22 414 l 22 414 l for volume and 273 15 K 273 15 K for temperature into the ideal gas equation and solve for r

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web chemists sometimes make comparisons against a standard temperature and pressure (STP) for reporting properties of gases 273 15 K and 1 atm 101 325 kPa 1 atm STP one mole of an ideal gas has a volume of about 22.4 l this is referred to as the standard molar volume figure 9 18

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web to separate the lanthanides from other elements occurring with them they are chemically combined with specific substances to form lanthanide compounds with low solubility oxalates and fluorides for example a process known as ion exchange is then used to separate the lanthanides from each other in this process a solution of the

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