

# Stoichiometry

Solve the following problems:

1. Four moles of iron (II) sulfide added to seven moles of oxygen, produced two moles of iron (III) oxide and four moles of sulfur dioxide. (20pts)

a) Translate the problem to chemical equation.



**Reactant side**

4 Fe atoms

4 S atoms

7(2) = 14 O atoms

**Product side**

4 Fe atoms

4 S atoms

2(3) + 4(2) = 14 O atoms

b) How many grams of oxygen will react with 2.0 grams of iron (II) sulfide?

**Given:**

m FeS = 2.0 g

Req'd. = m O<sub>2</sub>

**Solution:**

molar mass of FeS = 88 g/mol

mole ratio = 4 mol FeS: 7 mol O<sub>2</sub>

molar mass of O<sub>2</sub> = 32 g/mol

$$m_{O_2} = 2.0 \cancel{g_{FeS}} \left( \frac{1 \cancel{mol_{FeS}}}{88 \cancel{g_{FeS}}} \right) \left( \frac{7 \cancel{mol_{O_2}}}{4 \cancel{mol_{FeS}}} \right) \left( \frac{32 \cancel{g_{O_2}}}{1 \cancel{mol_{O_2}}} \right) = 1.3 g_{O_2}$$

2. Rusting involves the reaction of iron, Fe with oxygen, O<sub>2</sub> to form iron (III) oxide, Fe<sub>2</sub>O<sub>3</sub>. (20pts)

a) Write the balanced chemical equation.



**Reactant side**

4 Fe atoms

3(2) = 6 O atoms

**Product side**

4 Fe atoms

2(3) = 6 O atoms

b) What are the mole ratios of the product to each reactant and of the reactants to each other?

**Reactant to each other:**

4 mol Fe: 3 mol O<sub>2</sub>

**Product to each reactant:**

2 mol Fe<sub>2</sub>O<sub>3</sub>: 4 mol Fe

# Answer Key To Stoichiometry Homework Problems

**H. Scott Fogler**



## **Answer Key To Stoichiometry Homework Problems:**

## **Answer Key To Stoichiometry Homework Problems** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Answer Key To Stoichiometry Homework Problems**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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