

Stoichiometry Worksheet 2
Chemistry

Name _____

Date _____ Hour _____

1. In the reaction : $2\text{Al}_2\text{O}_{3(g)} \rightarrow 4\text{Al}_{(s)} + 3\text{O}_{2(g)}$, the mole ratios are:

Al_2O_3 to Al _____ Al_2O_3 to O_2 _____ Al to O_2 _____ O_2 to Al _____

2. In the reaction: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, the mole ratio of oxygen to water is: _____

3. In the reaction: $\text{Ca} + \text{Cl}_2 \rightarrow \text{CaCl}_2$, the mole ratio of chlorine to calcium is: _____

4. In the reaction: $\text{Zn} + \text{H}_2\text{SO}_4 \rightarrow \text{ZnSO}_4 + \text{H}_2$, the mole ratio of zinc to sulfuric acid is:

5. The equation for the Haber process is represented by:



To convert 9.0 moles of hydrogen to ammonia would require how many moles of nitrogen?

6. In the equation: $2\text{KClO}_{3(g)} \rightarrow 2\text{KCl}_{(s)} + 3\text{O}_{2(g)}$, how many moles of oxygen are produced when 3.00 moles of KClO_3 decompose completely?

7. For the reaction: $\text{C}_{(s)} + 2\text{H}_{2(g)} \rightarrow \text{CH}_{4(g)}$, how many moles of H_2 are required to produce 10.0 moles of CH_4 ?

Answers Chemistry The Mole Review Questions

J Elliott



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Homework Helpers: Chemistry Greg Curran, 2025-09-12 Homework Helpers Chemistry is a user friendly review book that will make every student or parent trying to help their child feel like he or she has a private Chemistry tutor Concepts are explained in clear easy to understand language and problems are worked out with step by step methods that are easy to follow Each lesson comes with numerous review questions and answer keynotes that explain each correct answer and why it s correct This book covers all of the topics in a typical one year Chemistry curriculum including A systematic approach to problem solving conversions and the use of units Naming compounds writing formulas and balancing chemical equations Gas laws chemical kinetics acids and bases electrochemistry and more While Homework Helpers Chemistry is an excellent review for any standardized Chemistry test including the SAT II its real value is in providing support and guidance during the year s entire course of study

How to Master the BMAT Chris John Tyreman, 2011-05-03 The BMAT BioMedical Admissions Tests is a subject specific admissions test taken by applicants to medicine veterinary science and related courses at Cambridge Oxford Imperial College UCL and Royal Veterinary College With more students than ever achieving A grades at A level BMAT testing is becoming increasingly popular as a way for universities to identify the most able and dedicated applicants How to Master the BMAT consists of six sections of revision materials for the maths physics chemistry and biology components of the BMAT with additional notes for problem solving and the writing task At the end of each of these six sections a set of review questions enables you to check that you have the necessary core knowledge some of which you will have already acquired through your GCSE and A level studies The additional mock tests should be attempted under exam conditions for the best possible preparation They provide the equivalent of three full BMAT tests nine papers Whatever your knowledge base you are supported throughout the book and where possible each question is complete with revision topics indicated in brackets

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AFOSR Chemical & Atmospheric Sciences Program Review United States. Air Force. Directorate of Chemical and Atmospheric Sciences, 1972

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Chemistry James E. Brady, John R. Holum, 1995-12-29 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping readers develop skills in analytical thinking and problems solving

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Chemistry and Physics for Nurse Anesthesia,

Second Edition David Shubert, PhD, John Leyba, PhD, 2013-03-15 Print CourseSmart *Ebook: Chemistry* Julia Burdge, 2014-10-16 Chemistry Third Edition by Julia Burdge offers a clear writing style written with the students in mind. Julia uses her background of teaching hundreds of general chemistry students per year and creates content to offer more detailed explanation on areas where she knows they have problems. With outstanding art, a consistent problem solving approach, interesting applications woven throughout the chapters, and a wide range of end of chapter problems, this is a great third edition text.

Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter. Key revisions in this edition focus on three areas: The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry; Simultaneously, examples and questions have been updated to align them with career concepts relevant to the environmental, engineering, biological, pharmaceutical, and medical sciences; Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text. When students know what they know, they are better able to learn and incorporate the material. Providing a total solution through New WileyPLUS by fully integrating the enhanced eText with online assessment, answer specific responses, and additional practice resources. The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge. Problems are arranged in an intuitive confidence building order.

Student Solutions Manual Darrell Ebbing, 2005-03-17 Provides solutions to odd numbered Practice Problems, General Problems, and Cumulative Skills Problems, plus answers to Review Questions.

[Foundations of College Chemistry](#) Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students. It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence. The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need. This International Adaptation offers new and updated content with improved presentation of all course material. It builds on the strengths of previous editions, including clear explanations and step by step problem solving. The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry. There is new and expanded coverage of polarizing power and polarizability, Fajans rules, collision number, and mean free path, abnormal molecular masses, and van t Hoff factor, and applications of radioactivity.

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Jacaranda Chemistry 2 VCE Units 3 and 4, 3e learnON and Print Robert Stokes, Angela Stubbs, Neale Taylor, Jason Bourke, Ben Williams, Maida Derbogossian, 2023-12-26 Developed by expert Victorian teachers for VCE students The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023 2027

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