Analytical Chemistry MCQs: TYBSC SEM III

 The solubility product K_{sp} of sparingly soluble Ag₂CrO₄ is 4*10⁻¹². The molar solubility of the salt is

- (a) 1.0*10* moles/lit
- (b) 2.0*104 moles/lit
- (c) 1.0*10-5 moles/lit
- (d) 1.2*10-12 moles/lit

 Solubility product of AgCl is 1.44*10⁻⁴ moles/lit at 373K. The solubility of the AgCl is boiling of water will be....

- (a) 0.72*104 moles/lit
- (b) 0.72*10-2 moles/lit
- (c) 1.20*10-2 moles/lit
- (d)1.20*10⁻⁴ moles/lit
- 10. Precipitation takes place when the ionic product...
- (a) Is zero
- (b) Exceeds the solubilty product
- (c) Equal to solubilty product
- (d) Is less than the solubility product
- 11. The electomotive force is expressed In the unit of
- (a) Ohm
- (b) Volt
- (c) Ampere
- (d) Colulomb
- 12. If the cell can provide electrical energy to an external system, it is called as
- (a) Galvanic cell
- (b) Electrolytic cell
- (c) Concentration cell
- (d) Chemical cell
- The current per unit area of electrode surface is known as....
- (a) Current efficiency
- (b) Current yield
- (e) Current density
- (d) None of these

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Quality Control in Analytical Chemistry G. Kateman, L. Buydens, 1993-09-06 Describes the basics of analytical techniques sampling and data handling in order to improve quality control in analytical laboratory management Stresses what quality parameters can be improved and which ones should be rectified first This edition includes numerous modern methods and the latest developments in time proven techniques Basic Analytical Chemistry Mr. Rohit Manglik, 2024-05-18 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels

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