

Advance Study Assignment - Experiment 26, 2046 Lab

v1

Use this page to REPLACE the advance study assignment that appears at the end of Experiment #26 in your lab textbook.



When 5.00 mL of 0.0120 M $\text{Ba}(\text{NO}_3)_2$ (aq) are mixed with 5.00 mL of 0.0300 M KIO_3 , a white precipitate of $\text{Ba}(\text{IO}_3)_2$ forms, according to the equilibrium above.

1. How many moles of Ba^{2+} are initially present?
2. How many moles of IO_3^{-} are initially present?
3. What is the concentration of Ba^{2+} in the initial mixture? *You may wish to record your answer in the grid below.*
4. What is the concentration of IO_3^{-} in the initial mixture? *You may wish to record your answer in the grid below.*

In an absorbance spectrophotometer, the equilibrium solution from question #1 was analyzed and found to contain a concentration of iodate ion, $[\text{IO}_3^{-}]$, equal to 7.00×10^{-3} M. (This value is also recorded on the grid below).

5. What was the change in concentration of iodate when the solution equilibrated? *You may wish to record your answer in the grid below.*
6. What was the change in concentration of barium ion when the solution equilibrated? (HINT: What is the relationship between Ba^{2+} and IO_3^{-} in the reaction?) *You may wish to record your answer in the grid below.*

	$[\text{Ba}^{2+}]$	$[\text{IO}_3^{-}]$
Initial		
Δ		
Equilibrium		7.00×10^{-3} M

7. Use the data from above to determine the equilibrium concentration of barium ion. *You may wish to record your answer in the grid above.*
8. Use the completed grid to determine the value of the equilibrium constant, K_{sp} .

NOTE: I have provided the Equilibrium grid ("ICE" diagram) above to aid you in your calculations, but it is NOT necessary that you submit it with this Advance Study Assignment on Canvas. All you need to do to complete the assignment is provide answers for the 8 questions; the ICE diagram is just there for your convenience in organizing your calculations.

Advance Study Assignment Experiment 10 Answers

Jerry D. Wilson



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