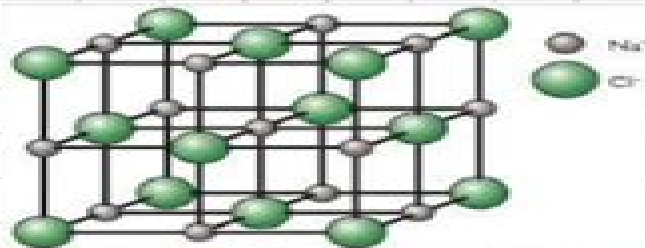


Ionic Bonding

(Between Metals and Non Metals : Group 1,2,6,7)

Ionic Bond: Bond formed by the electrostatic attraction between oppositely charged ions



Giant Lattice Structure
(exploded view)

EVIDENCE FOR THE EXISTENCE OF IONS

- 1) Electrolysis of aqueous ionic compound or a solution of it
- 2) Potassium Permanganate experiment

A) Experiment 1 : Electrolysis

- (a) • CuCrO_4 is dissolved in HCl acid. It is saturated with urea which increases the density of the solution.
- It is poured into a U tube, the liquid is originally green but as the electricity is applied the color shifts up the tube as the ions migrate towards the electrodes.

* $\text{CuCrO}_4 \rightarrow \text{Green}$

* -ve electrode $\rightarrow$ Blue Cu^{2+} ions

* +ve electrode $\rightarrow$ Yellow CrO_4^{2-} ions

- (b) • NaCl in aqueous form is poured into the U-tube and electricity is passed through it.

* Aqueous $\text{NaCl} \rightarrow$ Colorless

* -ve electrode $\Rightarrow$ Solid Na forms

* +ve electrode $\rightarrow$ Cl_2 gas is given off that can bleach damp litmus paper

Advanced Ionic Bonding Chem Quest 20

LL Leslie



Advanced Ionic Bonding Chem Quest 20:

Unveiling the Magic of Words: A Overview of "**Advanced Ionic Bonding Chem Quest 20**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Advanced Ionic Bonding Chem Quest 20**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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