13. (4 pts) What is an ionic compound?
    a. A compound that will dissolve in water
    b. Two nonmetal atoms bonded together
    c. Two metal atoms bonded together
    d. A metal and a nonmetal atom bonded together
    e. None of the above

14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
    a. Ionic
    b. Covalent
    c. Both a and b
    d. Neither a nor b

15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
    a. Ionic
    b. Covalent
    c. Both a and b
    d. Neither a nor b

16. (4 pts) How many total atoms are in this molecule: $C_6H_{12}OH$?
    a. 3
    b. 4
    c. 19
    d. 20
    e. None of the above

17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is $C_3H_{10}Cl_x$, what must x be?
    a. 1
    b. 2
    c. 3
    d. 4
    e. None of the above

18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water? C_3H_6ClF $C_6H_5CH_3$ CS_2 $BaSO_4$
    a. C_3H_6ClF
    b. $C_6H_5CH_3$
    c. CS_2
    d. $BaSO_4$
    e. CS_2 and $BaSO_4$

Answers For Chemistry Module 13

Raffaela Di Napoli

Answers For Chemistry Module 13:

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