Question 1: Water at 20° C falls from a height of 854 m. If the whole energy is used in increasing the temperature, find out the final temperature. Specific heat of water is 4200 J/K kg.

Given that

$$T_1 = 20^{\circ}C = 20 + 273 = 293 \,\mathrm{K}$$

height =  $h = 854 \,\mathrm{m}$ 

Specific heat of water = 4200 JK 'kg'

Final temperature =  $T_2 = ?$ 

P.E =  $mgh$  — O

Solution:

we know that AQ = cm AT -

From  $\bigcirc$  and  $\bigcirc$ CHAT = WAL C AT = gh C (T2-T, ) - gh T2 - T1 = gh T2 = gh + T1  $=\frac{9.8 \times 854}{4200} + 293$  $= \frac{8369.2}{4200} + 293 = 1.99 + 293$ T, = 294.99 K = 295-273 °C T, = 22°C © learningparadisehub.com

# **Answer The Following Questions About The Thermodynamics Of**

Y Pai

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Theoretical Chemistry from the Standpoint of Avogardro's Rule & Thermodynamics Walther Nernst,1895 Ice and Refrigeration ,1892 Gravity and the Quantum Jasjeet Singh Bagla, Sunu Engineer, 2017-02-23 This book provides a compilation of in depth articles and reviews on key topics within gravitation cosmology and related issues It is a celebratory volume dedicated to Prof Thanu Padmanabhan Paddy the renowned relativist and cosmologist from IUCAA India on the occasion of his 60th birthday The authors many of them leaders of their fields are all colleagues collaborators and former students of Paddy who have worked with him over a research career spanning more than four decades Paddy is a scientist of

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Sustainable Energy Jefferson W. Tester,2005 Evaluates trade offs and uncertainties inherent in achieving sustainable energy analyzes the major energy technologies and provides a framework for assessing policy options The Stability of Minerals G.D. Price,N.L. Ross,2007-11-23 30% discount for members of The Mineralogical Society of Britain and Ireland This volume addresses the fundamental factors that underlie our understanding of mineral behaviour and crystal chemistry a timely topic given current advances in research into the complex behaviour of solids and supercomputing Materials

Thermodynamics: With Emphasis On Chemical Approach (With Cd-rom) Hae-geon Lee,2012-02-28 This book is the expanded edition of the first book entitled Chemical Thermodynamics for Metals and Materials This new version presents thermodynamics of materials with emphasis on the chemical approach and is thus suitable for students in materials science and metallurgical engineering as well as related fields such as chemical engineering and physical chemistry

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Analytical Thermodynamics Dongqing Li,2024-12-07 This second edition presents an enriched and expanded exploration of the fundamental principles of thermodynamics tailored for graduate level studies Drawing on over three decades of academic teaching experience the author has refined the content making it more accessible and comprehensive Chapter 1 has been restructured for clarity delineating Legendre Transformation and Thermodynamic Potentials into separate sections while the treatment of Chemical Potentials has been significantly augmented encompassing two component ideal gas mixtures and a re derivation of chemical potentials for dilute solutions Additionally the section on thermodynamic stability now boasts enhanced explanations and illustrative figures Chapter 2 introduces a groundbreaking section Electrolyte

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