

Grade 11 Chemistry

INTERMOLECULAR FORCES & Physical Properties

EXAM QUESTIONS PART 3



NAME OF SUBSTANCE	FORMULA	BOILING POINT (°C)
Water	H ₂ O	100
Ethanol	CH ₃ CH ₂ OH	78
Bromine	Br ₂	58,8
Iodine	I ₂	184,3
Ammonia	NH ₃	-33,3
Phosphine	PH ₃	-87,7

Explain why ethanol is soluble in water. Refer to the relative strength of the intermolecular forces in ethanol and water.

Explain why the boiling point of iodine is higher than that of bromine. Refer to the intermolecular forces present in EACH substance in the explanation.

Answers For Grade 11 Intermolecular Formal Task

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Answeres For Grade 11 Intermolecular Formal Task:

This book delves into Answers For Grade 11 Intermolecular Formal Task. Answers For Grade 11 Intermolecular Formal Task is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Answers For Grade 11 Intermolecular Formal Task, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Answers For Grade 11 Intermolecular Formal Task
 - Chapter 2: Essential Elements of Answers For Grade 11 Intermolecular Formal Task
 - Chapter 3: Answers For Grade 11 Intermolecular Formal Task in Everyday Life
 - Chapter 4: Answers For Grade 11 Intermolecular Formal Task in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Answers For Grade 11 Intermolecular Formal Task. The first chapter will explore what Answers For Grade 11 Intermolecular Formal Task is, why Answers For Grade 11 Intermolecular Formal Task is vital, and how to effectively learn about Answers For Grade 11 Intermolecular Formal Task.
 3. In chapter 2, the author will delve into the foundational concepts of Answers For Grade 11 Intermolecular Formal Task. This chapter will elucidate the essential principles that must be understood to grasp Answers For Grade 11 Intermolecular Formal Task in its entirety.
 4. In chapter 3, this book will examine the practical applications of Answers For Grade 11 Intermolecular Formal Task in daily life. This chapter will showcase real-world examples of how Answers For Grade 11 Intermolecular Formal Task can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Answers For Grade 11 Intermolecular Formal Task in specific contexts. This chapter will explore how Answers For Grade 11 Intermolecular Formal Task is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Answers For Grade 11 Intermolecular Formal Task. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Answers For Grade 11 Intermolecular Formal Task.

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Answers For Grade 11 Intermolecular Formal Task Introduction

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