

TYPES OF CHEMICAL BONDS

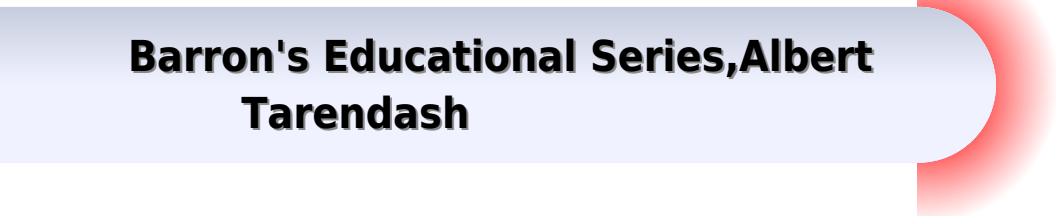
Name _____

Classify the following compounds as ionic (metal + nonmetal), covalent (nonmetal + nonmetal) or both (compound containing a polyatomic ion).

- | | | | |
|-----------------------------|-----------------|----------------------------|-----------------|
| 1. CaCl_2 | <u>ionic</u> | 11. MgO | <u>ionic</u> |
| 2. CO_2 | <u>Covalent</u> | 12. NH_4Cl | <u>both</u> |
| 3. H_2O | <u>Covalent</u> | 13. HCl | <u>Covalent</u> |
| 4. BaSO_4 | <u>both</u> | 14. KI | <u>ionic</u> |
| 5. K_2O | <u>ionic</u> | 15. NaOH | <u>both</u> |
| 6. NaF | <u>ionic</u> | 16. NO_2 | <u>Covalent</u> |
| 7. Na_2CO_3 | <u>both</u> | 17. AlPO_4 | <u>Both</u> |
| 8. CH_4 | <u>Covalent</u> | 18. FeCl_3 | <u>ionic</u> |
| 9. SO_3 | <u>Covalent</u> | 19. P_2O_5 | <u>Covalent</u> |
| 10. LiBr | <u>ionic</u> | 20. N_2O_5 | <u>Covalent</u> |

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