(d) The equilibrium mixture leaving the reaction chamber contains 15% ammonia. Suggest how the ammonia could be separated from the mixture.

	boiling point("C
hydrogen.	-253
mitrogen	~196
aurentoria.	-33

(w) Ammonia is used to make nitrogen trifluoride, NF_x.
Nitrogen trifluoride is essential to the electronics industry. It is made by the following reaction.

Determine if the above reaction is exothermic or endothermic using the following bond energies and by completing the following table. The first line has been done as an example. Bond energy is the amount of energy, in kJ/mole, needed to break or make one mole of the theoretic.

bond	bond energy in kJ./mole
M-H	390
F = F	155
M-E	280
14F	565

biomd	energy change/kJ
N=H	(3 - 390) - 1170
$F \rightarrow F$	
14-15	
H-F	

- 1943

[Total: 16]

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Zhanpeng Huang,Kun Wang,Jiandong Liu

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such as the short chain fatty acids SCFA producing bacteria which are known to promote immune tolerance Moreover the mucus layer contains anti microbial peptides AMPs and mucins that have key immune modulatory functions The integrated response of these combined defense systems is fundamental for containing microbes and their products within the intestine for avoiding their systemic spread and for suppressing their capacity to activate systemic immune and autoimmune responses

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