

Chemical 10

Misobers in Chemistry

Name: _____

Date: _____

Class: _____

Instructions: Qualitative vs. Quantitative

The following observations are qualitative

describing another fact.

It tells us things about how one substance reacts with another.
(It doesn't tell us how much)

The following observations are quantitative

because they tell us how much

Something reacts. All things given
in per cent or volume ratio.

Critical Thinking Questions

1. "What is the difference between qualitative and quantitative observations?" These questions should
elicit an understanding of the difference between qualitative and quantitative.
2. "What advantage of quantitative observations over qualitative ones does it have or not have?"
3. "What is a quantitative result in terms of chemical observations and chemistry?"

Instructions: Table

The following table contains quantitative and qualitative problems.

Table 1: Quantitative

Quantity	Unit	Unit Conversion
1. 100g	kg	100
1000g	kg	1000
1000g	kg	1000
1. 1000g	kg	1000
1. 1000g	kg	1000
1000g and 1000g	kg	1000

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