

| Element    | Atomic number | Electron configuration     |
|------------|---------------|----------------------------|
| hydrogen   | 1             | $1s^1$                     |
| helium     | 2             | $1s^2$                     |
| lithium    | 3             | $1s^2 2s^1$                |
| beryllium  | 4             | $1s^2 2s^2$                |
| boron      | 5             | $1s^2 2s^2 2p^1$           |
| carbon     | 6             | $1s^2 2s^2 2p^2$           |
| nitrogen   | 7             | $1s^2 2s^2 2p^3$           |
| oxygen     | 8             | $1s^2 2s^2 2p^4$           |
| fluorine   | 9             | $1s^2 2s^2 2p^5$           |
| neon       | 10            | $1s^2 2s^2 2p^6$           |
| sodium     | 11            | $1s^2 2s^2 2p^6 3s^1$      |
| magnesium  | 12            | $1s^2 2s^2 2p^6 3s^2$      |
| aluminum   | 13            | $1s^2 2s^2 2p^6 3s^2 3p^1$ |
| silicon    | 14            | $1s^2 2s^2 2p^6 3s^2 3p^2$ |
| phosphorus | 15            | $1s^2 2s^2 2p^6 3s^2 3p^3$ |
| sulfur     | 16            | $1s^2 2s^2 2p^6 3s^2 3p^4$ |
| chlorine   | 17            | $1s^2 2s^2 2p^6 3s^2 3p^5$ |
| argon      | 18            | $1s^2 2s^2 2p^6 3s^2 3p^6$ |

# 10 Electron Configurations Of Atoms And Ion Lab

**DP Hallahan**



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