

- (60) A helium cylinder in which the volume of gas = 2.24 L, at STP (1 atm, 273 K) developed a leak and when the leak was plugged the pressure in the cylinder was seen to have dropped to 550 mm of Hg. The number of moles of the gas that had escaped due to this leak is
 (A) 0.028 (B) 0.072 (C) 0.972 (D) 0.099
- (61) The anhydride of HNO_3 is
 (A) NO (B) NO_2 (C) N_2O (D) N_2O_5
- (62) An excess of aqueous ammonia is added to three different flasks (F_1 , F_2 , F_3) containing aqueous solutions of CuSO_4 , $\text{Fe}_2(\text{SO}_4)_3$ and NiSO_4 respectively. Which of the following is correct about this addition?
 I. A precipitate will be formed in all three flasks.
 II. Ammonia acts as a base as well as a ligand exchange reagent in F_1 and F_2 .
 III. A soluble complex of NH_3 and the metal ion is formed in F_1 and F_3 .
 IV. A precipitate will be formed only in F_2 .
 (A) I only (B) IV only (C) I and IV (D) II, III and IV
- (63) The mass (g) of NaCl that has to be dissolved to reduce the vapour pressure of 100 g of water by 10% (Molar mass of $\text{NaCl} = 58.5 \text{ g mol}^{-1}$) is
 (A) 26.11 g (B) 17.54 g (C) 81.25 g (D) 3.61 g
- (64) The reagent/s that can be used to separate norethindrone and norethanol from their mixture is/are



- (A) III (B) I and IV (C) I, II and III (D) II

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