

Page 3	Mark Scheme	Syllabus	Paper
	IGCSE – October/November 2013	0620	62
	(f) all points plotted correctly (4), –1 each incorrect straight line drawn with a ruler through all points except 0.4 g (1)		[5]
	(g) point at 0.4 g mass calcium / 0.21 g oxygen / Experiment 4 (1)		[1]
	(h) any evidence of extrapolation / indication (1) 0.45 g oxygen reacted (1) mass of calcium oxide = 1.15 g (1)		[3]
4	(c) table of results for Experiments 1, 2 and 3 initial temperature boxes completed correctly (1) 23, 22, 21 maximum temperature boxes correctly completed (1) 26, 24, 71 temperature rises correct (1) 3, 2, 50		[3]
	(e) table of results for Experiments 4 and 5 initial and maximum temperature boxes completed correctly (2) 19, 21 44, 29		[2]
	all temperature rises correct in tables (1) 25, 8		[1]
	(f) appropriate scale for y axis (1) bars inserted at correct heights (3) –1 for any incorrect not: a line graph labels (1)		[5]
	(g) (i) temperature rises greatest in Experiment 3 (1) (ii) magnesium is most reactive / more reactive (1)		[2]
	(h) hydrogen (1)		[1]
	(i) (i) copper (1) (ii) displacement / redox / exothermic (1) allow: oxidation / reduction		[2]
	(j) solid would react slower / temperature rises would be lower / less temperature change (1) smaller / less surface area (1) or same temperature (1) same mass of magnesium used (1)		[2]
	(k) dangerous / too reactive / explodes / owtte (1)		[1]

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Judd E. Hollander



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