

Page 3	Mark Scheme	Syllabus	Paper
	IGCSE – October/November 2013	0620	31
3	(a) (i) fractional distillation (liquid) air [1] [1] (ii) cracking / heat in presence of catalyst of alkane / petroleum to give an alkene and hydrogen [1] [1] [1] OR: electrolysis (1) named electrolyte (1) hydrogen at cathode (1) OR: from methane (1) react water / steam (1) heat catalyst (1) only ACCEPT: water with methane or electrolysis		
	(b) (i) the pair with both graphs correct is C [1] NOTE: mark (b)(i) independent of (b)(i) (ii) high pressure favours side with lower volume / fewer moles this is RHS / product / ammonia [1] [1] [1] %N₂ / yield increases as pressure increases. the forward reaction is exothermic; exothermic reactions favoured by low temperatures [1] [1] [1] %N₂ / yield decreases as temperature increases. ACCEPT: reverse arguments (iii) increases reaction rate [1] ACCEPT: reduces activation energy [1] OR: decreases the amount of energy particles need to react OR: economic rate at lower temperature so higher yield		
			[Total: 14]
4	(a) (i) (mass at t = 0) – (mass at t = 5) NOTE: must have mass at t = 5 not final mass [1] (ii) fastest at origin slowing down between origin and flat section gradient = 0 where gradient = 0 three of above in approximately the correct positions [2] (iii) 3 correct comments about gradient = [2] 2 correct comments about gradient = [1] 1 correct comment about gradient = [0]		
	(b) start at origin and smaller gradient same final mass just approximate rather than exact [1] [1]		

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