

Alkanes, Alkenes, Alkynes and Cyclic Hydrocarbons - Worksheet - ANSWERS

Name or draw the following compounds:

	Chemical structure	IUPAC Name
1	$\text{H}_3\text{C}-\text{CH}_2-\overset{\text{Cl}}{\underset{ }{\text{CH}}}-\text{CH}_3$	2-chlorobutane
2	$\begin{array}{c} \text{CH}_3 \\ \\ \text{H}_3\text{C}-\text{CH}-\text{CH}_2-\text{HC}-\text{CH}_3 \\ \quad \\ \text{CH}_3 \text{CH}_3 \end{array}$	3,5-dimethylheptane
3	$\text{H}_3\text{C}-\overset{\text{Br}}{\underset{ }{\text{CH}}}-\text{CH}_2-\overset{\text{Br}}{\underset{ }{\text{CH}}}-\underset{\text{CH}_3}{\underset{ }{\text{CH}}}-\underset{\text{Cl}}{\underset{ }{\text{C}}}-\text{CH}_3$	4,6-dibromo-2,2-dichloro-3-methylheptane
4	$\begin{array}{c} \text{CH}_3 \\ \\ \text{H}_3\text{C}-\text{C}-\text{CH}_2-\underset{\text{NH}_2}{\underset{ }{\text{CH}}}-\text{CH}_2-\text{CH}_2-\text{NO}_2 \\ \\ \text{CH}_3 \end{array}$	3-amino-5-ethyl-5-methyl-1-nitroheptane
5	$\text{H}_3\text{C}-\underset{\text{CH}_3}{\underset{ }{\text{CH}}}-\text{CH}_2-\underset{\text{CH}_3}{\underset{ }{\text{C}}}-\underset{\text{CH}_3}{\underset{ }{\text{CH}}}-\text{CH}_2-\text{CH}_3$	2,4,4-trimethylhexane
6	$\begin{array}{c} \text{CH}_3 \\ \\ \text{HC}-\text{CH}_2-\underset{\text{Cl}}{\underset{ }{\text{CH}}}-\underset{\text{Cl}}{\underset{ }{\text{CH}}}-\text{CH}_3 \\ \\ \text{CH}_2 \\ \\ \text{Cl}-\underset{\text{CH}_3}{\underset{ }{\text{CH}}} \end{array}$	2,3,7-trichloro-5-ethyloctane
7	$\text{H}_3\text{C}-\underset{\text{NO}_2}{\underset{ }{\text{C}}}(\text{NH}_2)-\text{CH}_2-\underset{\text{NO}_2}{\underset{ }{\text{C}}}(\text{NH}_2)-\text{CH}_3$	2,4-diamino-2,4-dinitropentane
8	$\text{Cl}-\underset{\text{Cl}}{\underset{ }{\text{C}}}(\text{Cl})-\text{CH}_2-\underset{\text{Cl}}{\underset{ }{\text{C}}}(\text{Cl})-\text{CH}_2-\underset{\text{Cl}}{\underset{ }{\text{C}}}(\text{Cl})-\text{CH}_3$	1,1,1,3,3,5,5-heptachlorohexane
9	$\text{H}_3\text{C}-\underset{\text{CH}_3}{\underset{ }{\text{C}}}(\text{CH}_3)-\text{CH}_2-\overset{\text{Br}}{\underset{ }{\text{CH}}}-\text{CH}_3$	2-bromo-4,4-dimethylhexane
10	$\text{H}_3\text{C}-\text{CH}_2-\underset{\text{Cl}}{\underset{ }{\text{C}}}(\text{Br})-\text{CH}_2-\text{CH}_3$	3-bromo-3-chloropentane
11	$\text{H}_3\text{C}-\underset{\text{NO}_2}{\underset{ }{\text{CH}}}-\text{CH}_2-\overset{\text{Br}}{\underset{ }{\text{CH}}}-\text{CH}_3$	1-bromo-3-nitrobutane
12	$\text{CH}_3-\overset{\text{Cl}}{\underset{ }{\text{CH}}}-\underset{\text{Cl}}{\underset{ }{\text{CH}}}-\text{CH}_2-\text{CH}_3$	2,3-dichloropentane

Alkenes And Alkynes Study Guide Answers

Karen Timberlake



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Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems Organic Chemistry, 13e Student Study Guide and Solutions Manual T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-05-03 Organic Chemistry Student Study Guide and Solutions Manual 13th Edition offers the full solutions for select exercises from the text **Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us **Organic Chemistry, Study Guide and Solutions Manual** T. W. Graham Solomons, Craig B. Fryhle, 2000 On the cover of this book is a Pacific yew tree found in the ancient forests of the Pacific Northwest The bark of the Pacific yew tree produces Taxol found to be a highly effective drug against ovarian and breast cancer Taxol blocks mitosis during eukaryotic cell division The supply of Taxol from the Pacific yew tree is vanishingly small however A single 100 year old tree provides only about one dose of the drug roughly 300 mg For this reason as well as the spectacular molecular architecture of Taxol synthetic organic chemists fiercely undertook efforts to synthesize it Five total syntheses of Taxol have thus far been reported Now a combination of isolation of a related metabolite from European yew needles and synthesis of Taxol from that intermediate supply the clinical demand This case clearly demonstrates the importance of synthesis and the use of organic chemistry It's just one of the many examples used in the text that will spark the interest of students and get them involved in the study of organic chemistry

Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011 *Study Guide and Solutions Manual for*

McMurry's Organic Chemistry, Fifth Edition Susan McMurry, 2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Price winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines *Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry* Katherine J. Denniston, Denniston Katherine, 2006 *Study Guide to Accompany Basics for Chemistry* Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students *Study Guide and Solutions Manual to Accompany The Essence of Organic Chemistry* Jane M. Cram, Donald J. Cram, 1978 *Organic Chemistry Study Guide* Robert J. Ouellette, J. David Rawn, 2014-11-04 Organic Chemistry Study Guide Key Concepts Problems and Solutions features hundreds of problems from the companion book Organic Chemistry and includes solutions for every problem Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject Organic chemistry is a constantly evolving field that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry springs from organic chemistry Additionally all scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic chemistry Organic chemistry like any skill is best learned

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