

Experiment 2 - Gravimetric Analysis of a Chloride Salt (G)

Purpose:

Determining quantitative value of chloride in an unknown soluble salt by showing and explaining common methods in gravimetric analysis.

Theory:

- Reaction between silver ion and chloride ion to form silver chloride as an equation:



- AgCl solubility in water and the K_{sp} :

Silver chloride is very insoluble, so we can assume that the above reaction goes to completion.

$$K_{sp} = 1.6 \times 10^{-10}$$

- AgCl is precipitated with an acidic solution to reduce likelihood of acid ions interfering with results. These ions will form co-precipitates with silver in the acid, due to these co-precipitates, the results would be higher.
- To prevent the precipitate from going through the filtered paper, it is heated to coagulate it into a different texture that's easier to work with. Heating the precipitate makes it coagulate, therefore, it is harder for the precipitate to go through the filtered paper.
- Dry silver chloride decomposes into silver and chlorine in the presence of light:



- If photodecomposition takes place in air, results will be lower, since chlorine will be released in the form of gas. If photochemical decomposition occurs in the presence of excess silver ions in aqueous solution, it will result in higher results.

Analysis Of An Unknown Salt Lab Report

Arturo Cuomo



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Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

Laboratory Experiments for General Chemistry Harold R. Hunt, Toby F. Block, 1994 **English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента** Надежда

Зорина, Александра Соболева, 2022-10-19 Scientific and Technical Aerospace Reports, 1995-05 *General Chemistry* Hunt, 1976-06-07 Laboratory Manual for Principles of General Chemistry J. A. Beran, Mark Lassiter, 2022-08-16

Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don't write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section

General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems

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1998 Report on Attainment of the California Particulate Matter Standards in the Monterey Bay Region Janet

Brennan, 1999 *U.S. Government Research Reports*, 1964 Analytical Chemistry for Technicians, Fourth Edition John Kenkel, 2013-08-13 Written as a training manual for chemistry based laboratory technicians this thoroughly updated fourth edition of the bestselling Analytical Chemistry for Technicians emphasizes the applied aspects rather than the theoretical ones The book begins with classical quantitative analysis and follows with a practical approach to the complex world of sophisticated electronic instrumentation commonly used in real world laboratories Providing a foundation for the two key qualities the analytical mindset and a basic understanding of the analytical instrumentation this book helps prepare individuals for success on the job Chapters cover sample preparation gravimetric analysis titrimetric analysis instrumental analysis spectrochemical methods such as atomic spectroscopy and UV Vis and IR molecular spectrometry chromatographic techniques including gas chromatography and high performance liquid chromatography electroanalytical methods and more Incorporating an additional ten years of teaching experience since the publication of the third edition the author has made significant updates and enhancements to the fourth edition More than 150 new photographs and either new or reworked drawings spanning every chapter to assist the visual learner A new chapter on mass spectrometry covering GC MS LC MS LC MS MS and ICP MS Thirteen new laboratory experiments An introductory section before chapter 1 to give students a preview of general laboratory considerations safety laboratory notebooks and instrumental analysis Additional end of chapter problems expanded report type questions and inclusion of relevant section headings in the Questions and Problems sections Application Notes in each chapter An appendix providing a glossary of quality assurance and good laboratory practice GLP terms **Fundamentals of Inorganic Chemistry** Bruce E. Wilcox, 1997-07 **Chemistry** Karen Timberlake, 1999 Suitable for one or two term lab courses covering general organic and biological chemistry this new edition written by Karen Timberlake features many improvements to the insightful experiments that have made it the leading lab manual Each experiment encourages critical thinking with laboratory goals discussion of related concepts clear instructions new pre lab questions and comprehensive report pages Forty one experiments illustrate the basic principles of chemistry **Report summaries** United States. Environmental Protection Agency, 1983 **Science Educator's Guide to Laboratory Assessment** Rodney L. Doran, 2002 The book opens with an up to date discussion of assessment theory research and uses Then comes a wealth of sample assessment activities in biology chemistry physics and Earth science Keyed to the National Science Education Standards the activities include reproducible task sheets and scoring rubrics All are ideal for helping

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web what are the differences between whales and dolphins whales and dolphins are both cetaceans but there are some differences between them whales are generally larger than dolphins and have a more streamlined body shape they also have a more complex social structure and tend to be more solitary

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web aug 26 2023 dolphin any of the toothed whales belonging to the mammal family delphinidae oceanic dolphins as well as the families platanistidae and iniidae the two that contain the river dolphins of the nearly 40 species of dolphins in the delphinidae 6 are commonly called whales including the killer whale and the pilot whales

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web feb 9 2022 whales are larger than dolphins in the vast majority of cases the only dolphin that can match the size of whales is the killer whale but that is actually a member of the dolphin family the average dolphin can reach a weight of 660lbs and grows about 13ft long but the orca can reach lengths of 26ft and a weight of 19 000lbs

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web whales and dolphins are mammals that belong to the order cetacea which also includes porpoises dolphins are a type of toothed whale while popular culture often celebrates the intelligence of dolphins whales and dolphins are generally thought to be equally intelligent according to recent scientific studies

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web usually found alone or in small groups do not echolocate did you know rorqual is derived from the norwegian word røykval which means furrow whale this is thought to refer to the long folds of skin in the lower jaw right whale and bowhead whale family

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the new zealand dolphin and tucuxi other ocean dolphins include risso s striped spinner spotted common dusky hourglass rough toothed white beaked commerson s and bottlenose dolphins

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web whales and dolphins belong to the order known as cetacea cetaceans consists of three groups of marine mammals which include whales dolphins and porpoises all three species share many of the same characteristics as land mammals including the need to breathe air being warm blooded giving birth producing milk having hair not all species

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web jul 21 2021 both animals can produce vocalizations to communicate but since whales communicate mostly underwater you might not be as familiar with their sounds as compared to those distinct clicks or whistles more commonly heard from dolphins dolphins also have the ability to echolocate where baleen whales do not food for thought

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web facts about whales and dolphins whale and dolphin conservation watch on top ten facts about dolphins there are currently 42 species of dolphins and seven species of porpoises dolphins are marine mammals they must surface to breathe air and give birth to live young a dolphin pregnancy last between nine and 16 months

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