# CHM 111 Analysis of Aluminum-Zinc Alloy using Ideal Gas Law

# Report Sheet

# Data Sheet:

	Trial 1	Trial 2	
Mass of gelatin capsule	1.7382 g		_ g
Mass of alloy sample plus capsule	1.9852 g		_ g
Mass of empty beaker	104.3024 g		g
Mass of beaker plus displaced water	346.225g		g
Barometric pressure		m Hg	
Temperature	_24	_°C	
Calculations Mass of alloy sample	0.247 g		_ g
Mass of displaced water	241.9226 g		g
Volume of displaced water	mL		mL
Volume of H <sub>2</sub> , V	L		L
Temperature of H <sub>2</sub> , T	297.15	_ K	
Vapor pressure of water at T	22.4	_ mm Hg	
Pressure of dry H2,	755.7 mm Hg;	0.9943 atm	
Moles H2 from sample,	moles		moles
Moles H2 per gram of sample,	moles/g		moles/g %
Al (read from graph)	%	m 41g	%
Unknown no5	Access to the second se		The second

# **Analysis Of Aluminum Zinc Alloy Lab Answers**

**Anthony S. Fauci** 

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