

Oxidation-Reduction Worksheet

For each reaction below, identify the atom oxidized, the atom reduced, the oxidizing agent, the reducing agent, the oxidation half reaction, the reduction half reaction, and then balance the equation by the method of oxidation-reduction showing all electrons transfers.

1. $\overset{0}{\text{Mg}} + 2\overset{+1}{\text{H}}\overset{-1}{\text{Cl}} \rightarrow \overset{+2}{\text{Mg}}\overset{-2}{\text{Cl}}_2 + \overset{0}{\text{H}}_2$ $\overset{0}{\text{Mg}} \xrightarrow{-2e^-} \overset{+2}{\text{Mg}}$ $2\overset{+1}{\text{H}} \xrightarrow{+2e^-} \overset{0}{\text{H}}_2$
2. $2\overset{0}{\text{Fe}} + 3\overset{+3}{\text{V}}_2\overset{-2}{\text{O}}_3 \rightarrow \overset{+3}{\text{Fe}}_2\overset{-2}{\text{O}}_3 + 6\overset{+2}{\text{V}}\overset{-2}{\text{O}}$ $2\overset{0}{\text{Fe}} \xrightarrow{-6e^-} \overset{+3}{\text{Fe}}_2$ $3\overset{+3}{\text{V}}_2 \xrightarrow{+6e^-} 6\overset{+2}{\text{V}}$
3. $\overset{+7}{\text{K}}\overset{+4}{\text{Mn}}\overset{-2}{\text{O}}_4 + \overset{+5}{\text{K}}\overset{-2}{\text{N}}\overset{-2}{\text{O}}_2 + 2\overset{+2}{\text{H}}_2\overset{-2}{\text{S}}\overset{-2}{\text{O}}_4 \rightarrow \overset{+2}{\text{Mn}}\overset{-2}{\text{S}}\overset{-2}{\text{O}}_4 + 2\overset{+2}{\text{H}}_2\overset{-2}{\text{O}} + \overset{+5}{\text{K}}\overset{-2}{\text{N}}\overset{-2}{\text{O}}_3 + \overset{+6}{\text{K}}_2\overset{-2}{\text{S}}\overset{-2}{\text{O}}_4$
4. $\text{K}_2\text{Cr}_2\text{O}_7 + \text{SnCl}_2 + \text{HCl} \rightarrow \text{CrCl}_3 + \text{SnCl}_4 + \text{KCl} + \text{H}_2\text{O}$
5. $\text{KMnO}_4 + \text{NaCl} + \text{H}_2\text{SO}_4 \rightarrow \text{Cl}_2 + \text{K}_2\text{SO}_4 + \text{MnSO}_4 + \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$
6. $\text{K}_2\text{Cr}_2\text{O}_7 + \text{H}_2\text{O} + \text{S} \rightarrow \text{SO}_2 + \text{KOH} + \text{Cr}_2\text{O}_3$
7. $\text{KClO}_3 + \text{C}_{12}\text{H}_{22}\text{O}_{11} \rightarrow \text{KCl} + \text{H}_2\text{O} + \text{CO}_2$
8. $\text{H}_2\text{C}_2\text{O}_4 + \text{K}_2\text{MnO}_4 \rightarrow \text{CO}_2 + \text{K}_2\text{O} + \text{Mn}_2\text{O}_3 + \text{H}_2\text{O}$
9. $\text{Mn}(\text{NO}_3)_2 + \text{NaBiO}_3 + \text{HNO}_3 \rightarrow \text{HMnO}_4 + \text{Bi}(\text{NO}_3)_3 + \text{NaNO}_3 + \text{H}_2\text{O}$
10. $\text{H}_2\text{C}_2\text{O}_4 + \text{KMnO}_4 \rightarrow \text{CO}_2 + \text{K}_2\text{O} + \text{Mn}_2\text{O}_3 + \text{H}_2\text{O}$

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monoethylene glycol MEG The process examined is similar to the Avantium YXY process followed by a process that copolymerizes FDCA and MEG In this process a 70 wt% glucose water syrup feed is isomerized to fructose Then fructose is converted to methoxy methyl furan MMF intermediate which is oxidized to FDCA This process generates methyl acetate and methyl levulinate as by products The FDCA generated and MEG are polymerized to PEF in two polymerization steps in order to obtain Bottle Grade PEF This report was developed based essentially on the following reference s 1 US Patent 8519167 issued to Furanix Technologies B V in 2013 2 US Patent 20160024039 issued to Furanix Technologies B V in 2016 3 WO Patent 2015137807 issued to Furanix Technologies in 2015 Keywords Polyethylene Furanoate PEF Green FDCA Hydroxyl Methyl Furfural HMF Purified Terephthalic Acid Polyethylene 2 5 Furandicarboxylate MEG **Polyethylene Furanoate Production from Raw Sugar - Cost Analysis - PEF E31B** Intratec,2019-09-17 This report presents a cost analysis of Polyethylene Furanoate PEF production starting from raw sugar and monoethylene glycol MEG The process examined is similar to the Avantium YXY process followed by a process that copolymerizes FDCA and MEG In this process a raw sugar sucrose is inverted to glucose and fructose The glucose is isomerized to fructose Then fructose is converted to methoxy methyl furan MMF intermediate which is oxidized to FDCA This process generates methyl acetate and methyl levulinate as by products The FDCA generated and MEG are polymerized to PEF in two polymerization steps in order to obtain Bottle Grade PEF This report was developed based essentially on the following reference s 1 US Patent 8519167 issued to Furanix Technologies B V in 2013 2 US Patent 20160024039 issued to Furanix Technologies B V in 2016 3 WO Patent 2015137807 issued to Furanix Technologies in 2015 Keywords Polyethylene Furanoate PEF Green FDCA Hydroxyl Methyl Furfural HMF Purified Terephthalic Acid Polyethylene 2 5 Furandicarboxylate MEG **Oswaal CDS Question Bank | Previous Years Solved Question Papers Chapter-Wise & Topic-Wise General Knowledge (2014-2023) For 2024 Exam** Oswaal Editorial Board,2024-01-19 Description of the product 100% updated with Fully Solved April September 2023 Papers Concept Clarity with detailed explanations of 2014 to 2023 Papers Extensive Practice with 1200 Questions and Two Sample Question Papers Crisp Revision with Concept Based Revision Notes Mind Maps Mnemonics Expert Tips helps you get expert knowledge master crack CDS in first attempt Exam insights with 5 Year wise 2019 2023 Trend Analysis empowering students to be 100% exam ready **Handbook of Water Analysis** Leo M.L. Nollet,Leen S. P. De Gelder,2000-06-27 This work details water sampling and preservation methods by enumerating the different ways to measure physical chemical organoleptical and radiological characteristics It provides step by step descriptions of separation residue determination and cleanup techniques for a variety of fresh and salt waters It also discusses information regarding the analysis and detection of bacteria and algae **MCAT Organic Chemistry Review 2025-2026** Kaplan Test Prep,2024-07-02 Kaplan s MCAT Organic Chemistry Review 2024 2025 offers an expert study plan detailed subject review and hundreds of online and in book practice questions all authored by the experts behind Kaplan s score raising MCAT prep course Prepping for the MCAT is a

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Chemical Engineering Design and Analysis T. Michael Duncan,Jeffrey A. Reimer,2019-01-24 The go to guide to learn the principles and practices of design and analysis in chemical engineering

Introductory Titrimetric and Gravimetric Analysis Evelyn M. Rattenbury,2016-06-06 Introductory Titrimetric and Gravimetric Analysis discusses the different types of titration and the weighing of different solutions in solid form Coverage is made on acid base titration argentometric titrations and oxidation reduction titrations Iodometric titrations and complexometric titrations are also explained Extensive discussion on each of the titration method along with some examples and laboratory experiments is given The process of weight measurement of damp powder is one example of the experiments The book is a manual that guides a student to the correct ways of conducting an experiment made on such solutions as sodium hydroxide using hydrochloric acid and oxalic acid Outcome of such experiments in terms of composition weight of solutions and measurement of pressure in certain environment is tabulated and briefly explained Logarithms and antilogarithms are included at the end of the book The text will serve as a good laboratory manual for students preparing for science examination as well as for chemists and chemical engineers

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