

Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON-METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.) H_2 hydrogen is diatomic



Single Bond

2.) F_2 fluorine is diatomic



3.) O_2 oxygen is diatomic



Double Bond

4.) N_2 nitrogen is diatomic. Is this a triple bond?



5.) BF_3 you need 3 fluorine atoms here



6.) Ammonia NH_3 hint: how many hydrogen atoms are needed?



7.) Carbon dioxide CO_2



8.) Methane CH_4 careful here 4 hydrogen atoms needed



9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!



10.) SO_2 hint: one pair of electrons from sulfur must be slit up for this one to work.



Answer Key To The Nature Of Covalent Bonding

D Kirk



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Theoretical Chemistry Accounts Christopher Cramer, D.G. Truhlar, 2013-11-11 For the New Century Issue of the journal Theoretical Chemistry Accounts the advisory editors identified papers from the first century of theoretical chemistry and discussed their importance for the twentieth century with an eye towards the twenty first century Sixty six such perspectives are published in the New Century Issue To make this unique collection available to younger scientists for entertaining reading and re reading of the original publications the publisher decided to reprint a special edition of the issue

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