



Equations

Name _____

Date _____

Write a balanced chemical equation to represent each of the following chemical reactions.

1. iron + sulfur $\rightarrow$ iron(II) sulfide
2. zinc + copper(II) sulfate $\rightarrow$ zinc sulfate + copper
3. silver nitrate + sodium bromide $\rightarrow$ sodium nitrate + silver bromide
4. potassium chlorate (heated) $\rightarrow$ potassium chloride + oxygen
5. water (electricity) $\rightarrow$ hydrogen + oxygen
6. mercury(II) oxide (heated) $\rightarrow$ mercury + oxygen
7. potassium iodide + lead(II) nitrate $\rightarrow$ lead(II) iodide + potassium nitrate
8. aluminum + oxygen $\rightarrow$ aluminum oxide
9. magnesium chloride + ammonium nitrate $\rightarrow$ magnesium nitrate + ammonium chloride
10. iron(III) chloride + ammonium hydroxide $\rightarrow$ iron(III) hydroxide + ammonium chloride
11. sodium peroxide + water $\rightarrow$ sodium hydroxide + oxygen
12. iron(III) oxide + carbon $\rightarrow$ iron + carbon monoxide
13. iron + water $\rightarrow$ hydrogen + iron(III) oxide
14. iron(III) chloride + potassium hydroxide $\rightarrow$ potassium chloride + iron(III) hydroxide
15. aluminum + sulfuric acid $\rightarrow$ aluminum sulfate + hydrogen
16. sodium carbonate + calcium hydroxide $\rightarrow$ sodium hydroxide + calcium carbonate



Answer Key Weston Walch Chemistry

Brian Pressley



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Table of Contents Answer Key Weston Walch Chemistry

1. Understanding the eBook Answer Key Weston Walch Chemistry
 - The Rise of Digital Reading Answer Key Weston Walch Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key Weston Walch Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key Weston Walch Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key Weston Walch Chemistry
 - Personalized Recommendations
 - Answer Key Weston Walch Chemistry User Reviews and Ratings
 - Answer Key Weston Walch Chemistry and Bestseller Lists
5. Accessing Answer Key Weston Walch Chemistry Free and Paid eBooks
 - Answer Key Weston Walch Chemistry Public Domain eBooks
 - Answer Key Weston Walch Chemistry eBook Subscription Services
 - Answer Key Weston Walch Chemistry Budget-Friendly Options
6. Navigating Answer Key Weston Walch Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key Weston Walch Chemistry Compatibility with Devices
 - Answer Key Weston Walch Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key Weston Walch Chemistry
 - Highlighting and Note-Taking Answer Key Weston Walch Chemistry
 - Interactive Elements Answer Key Weston Walch Chemistry
8. Staying Engaged with Answer Key Weston Walch Chemistry

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Answer Key Weston Walch Chemistry
- 9. Balancing eBooks and Physical Books Answer Key Weston Walch Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key Weston Walch Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key Weston Walch Chemistry
 - Setting Reading Goals Answer Key Weston Walch Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key Weston Walch Chemistry
 - Fact-Checking eBook Content of Answer Key Weston Walch Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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