## Name the following acids

1. HCL	hydrochloric acid
2. HClO <sub>4</sub>	perchloric acid
<ol> <li>HIO<sub>3</sub></li> </ol>	iodic acid
4. HII	hydroiodic acid
5. H <sub>2</sub> SO <sub>4</sub>	sulfuric acid
<ol> <li>H<sub>2</sub>S</li> </ol>	hydrosulfuric acid
7. HCN	hydrocyanic acid
<ol> <li>H<sub>2</sub>CO<sub>3</sub></li> </ol>	carbonic acid
9. HBrO <sub>4</sub>	perbromic acid
10.HBrO <sub>3</sub>	bromic acid
11.HC <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	acetic acid
12.H <sub>3</sub> PO <sub>3</sub>	phosphorous acid
13.H <sub>3</sub> P	hydrophosphoric acid
14.H <sub>2</sub> CrO <sub>4</sub>	chromic acid
15.H <sub>2</sub> CrO <sub>2</sub>	hypochromous acid
16.H <sub>2</sub> Cr	hydrochromic acid

## Give formulas for the following acids

17.hydrofluoric acid	HF
18.nitric acid	HNO <sub>3</sub>
19.nitrous acid	HNO <sub>2</sub>
20.hydroxic acid	H <sub>2</sub> O
21.periodic acid	HIO <sub>4</sub>
22. percarbonic acid	H <sub>2</sub> CO <sub>4</sub>
23.hypobromous acid	HBrO
24. bromous acid	HBrO <sub>2</sub>
25.permanganic acid	HMnO <sub>4</sub>
26.manganic acid	HMnO <sub>3</sub>
27. iodous acid	HIO <sub>2</sub>
28. sulfurous acid	H <sub>2</sub> SO <sub>3</sub>
29. sulfuric acid	H <sub>2</sub> SO <sub>4</sub>
30.perchloric acid	HCIO <sub>4</sub>
31.chlorous acid	HClO <sub>2</sub>
32.acetic acid	$HC_2H_2O_2$
	H <sub>3</sub> PO <sub>4</sub>
34. phosphorous acid	H <sub>3</sub> PO <sub>3</sub>

# **Answers For Naming Acids**

Barron's Educational Series, Neil D. Jespersen, Pamela Kerrigan

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