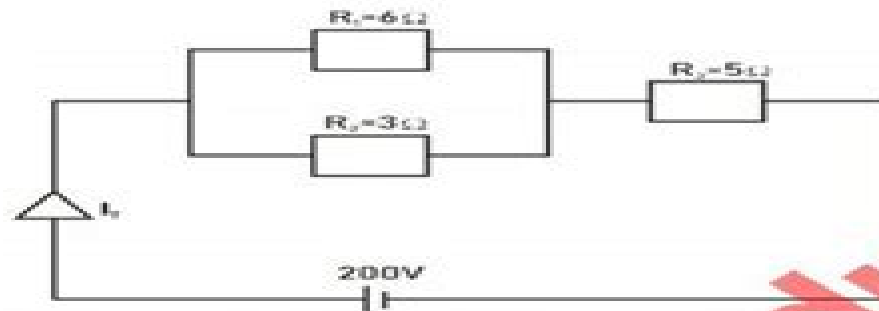


PAPER II

Answer any **four** questions.
All questions carry equal marks.

- 1(a) Calculate the total current in the circuit below



- (b) State **three** methods used for soldering a joint.
- 2(a) State **three** advantages of an aluminum conduct over steel conduct.
- (b) Mention any three points to be considered while determining the size of a conductor for wiring.
- (c) Calculate the resistance of a copper cable 100 m long, if it has cross-sectional area of 50 mm². The resistivity of a copper is 1.7 Ω /m.
- 3(a) Mention two differences between lap and wave windings.
- (b) List any five causes of accident in the workshop.
- (c) Explain the function of a carbon brush used in electric motor.
- 4(a) Define the term illumination.
- (b) A coil of insulated wire of resistance 8 Ω and inductance 0.03 H is connected to an a.c supply at 240 V, 50 Hz.
Calculate:
(i) current and
(ii) the power.
- (c) (i) briefly describe a Trunking System of wiring.
(ii) What is the major advantage a trunking has over conduit system.

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web write at least three conclusions from your exploration 1 protons and neutrons determine the mass of the atom 2 only protons determine the atomic number 3 the more unstable the atom the less presence or no presence in nature 1 protons and neutrons determine the mass of the atom 2 only protons determine the atomic number 3

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web microsoft word phet isotopes and atomic mass author lauralt2 created date 9 28 2017 9 36 15 am

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web isotopes and atomic mass model 1 make isotopes open the isotopes and atomic mass simulation play with the make isotopes tab of the simulation for a few minutes and then answer the following questions 1 what particles determine the mass number number of protons and neutrons 2 why is mass number always a whole

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web this guided inquiry activity has two learning goals students will be able to students will be able to 1 explain the difference between atomic mass and mass number and 2 calculate average atomic mass from percent abundance and isotopic mass

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web find the average atomic mass of an element given the abundance and mass of its isotopes predict how the mass and name of an isotope will change given a change in the number of protons neutrons or electrons

phet simulator isotopes and atomic mass guided inquiry - May 12 2023

web oct 22 2020 this a screencast video tutorial on how to answer and navigate the phet simulator named isotopes and atomic mass found at phet colorado edu sims ht