HOMOGENEOUS AQUEOUS SYSTEMS

Section Review

Objectives

- · Distinguish between a solvent and a solute
- · Describe what happens in the solution process
- Explain why all ionic compounds are electrolytes
- Demonstrate how to write the formula for a hydrate

Vocabulary

- aqueous solution
- solvent.
- solute

- solvation.
- electrolyte
- monelectrolyte

- strong electrolyte
- weak electrolyte
- hydrate.

Key Equation

Percent H₂O = mass of water x 100%

mass of hydrate

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Water is a polar liquid and an excellent ______ for many substances. Aqueous solutions are _____ mixtures of ions or molecules in water. The solubility of a solute depends on solute-

solvent interactions. A good rule to remember is ____3__.

Substances that dissolve as ions are known as __4___ A
solute that is completely ionized in solution is a __5__
electrolyte. A weak electrolyte is only __6__ ionized. A solution
of an electrolyte will __7__ an electric current, whereas a

solution of a ____8 __ is nonconducting.

Many crystals are ______; they contain water of hydration.

In the process called ______ the water of hydration is lost from a

hydrate that is exposed to air.

3.

4.

5. _____

6.

7.

8. ____

10

Chapter 15 Water and Aqueous Systems 379

Answers For Homogeneous Aqueous Systems Section Review

Albert A Gayle

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Applied Mechanics Reviews ,1974 Review of American Chemical Research William Albert Noves, 1898 Energy Research Abstracts ,1983 Advanced Oxidation Processes in Dye-Containing Wastewater Subramanian Senthilkannan Muthu, Ali Khadir, 2022-05-11 Among various industries releasing wastewater into the environment printing dyeing and textile industries are of great importance as they frequently contain high amounts of colorful compounds having high chemical and biological oxygen demands Health related effects of colorants are extensively reported which necessitates the seriousness of dye removal from water and wastewater The utilization of advanced oxidation processes AOPs in dve degradation has gained considerable attention recently due to the release of high energetic radicals as oxidants that are capable of removing dye compounds This Volume 1 presents versatile applications of AOPs in dye removal Accordingly processes such as Ozone based AOPs UV irradiation catalytic AOPs etc are discussed with the aim of dye removal under different operational parameters. The role of different nanoparticles is also investigated By presenting the fundamentals of AOPs as well as recent advances this book is useful for environmental engineers and chemists who are concerned with wastewater pollution and treatment **Nuclear Science Abstracts** ,1974 The Supramolecular Chemistry of Organic-Inorganic Hybrid Materials Knut Rurack, Ramon Martinez-Manez, 2010-04-07 The combination of supramolecular chemistry inorganic solids and nanotechnology has already led to significant advances in many areas such as sensing controlled motion and delivery By making possible an unprecedented tunability of the properties of nanomaterials these techniques open up whole new areas of application for future supramolecular concepts The Supramolecular Chemistry of Organic Inorganic Hybrid Materials gathers current knowledge on the subject and provides an overview of the present state and upcoming challenges in this rapidly growing highly cross or interdisciplinary research field The book details how these designed materials can improve existing materials or generate novel functional features such as chemical amplification cooperative binding and signal enhancement that are difficult or not at all achievable by classical organic supramolecular chemistry It also discusses issues related to nanofabrication or nanotechnology such as the directed and controlled assembly or disassembly biomimetic functions and strategies and the gating and switching of surface functions or morphology

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