

Day 1

Student Objectives

- · Read, spell, and write words with long a spelled at, a-consonant-e, or et.
- Read, spell, and write words with long e. spelled ee, ca, e-consonant-e, v, or i-consonant-e.

Unit Pretest

- Use pretest sentences on page T14A.
- To differentiate instruction, use pretest. results to assign students to the appropriate list and daily activities (see p. Z23).

Teaching Tips

WORD SORT Cards

- . Conduct a Teacher-Led Sort (see p. Z24).
- · Point out the patterns: The long a sound can be spelled al, a-consonant-e, or ei. The long e sound can be spelled ee, ea, e-consonant-e, y, or i-consonant-e. Have students find an example of each.
- Tell students that long e spelled ie is often confused with the ei spelling (e.g., neither, ceiling), which is used only 1% of the time. Hence the nunemonic: "I before e except after c or when sounded like a, as in neighbor or weigh."

Spelling Tutor

Have students use Spelling Tutor to study the spelling words. QR codes are also available at resources.zaner-bloser.com/spcn. (6)

Differentiated Word Lists



Home/School Spelling Proclice Moster*, TRB, p. 7





[&]quot;Available in English and Spanish

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Connections to THINKING

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Unusual entireds populate the island.

Think & Sort the spelling words.

Words can have more than one wavel sound. Focus on long a and Named at

1-2. Write the words that have both a long a and a long a sound

3-10. While the words that have the long a sound.

11-20. Write the words that have the long a sound.

Remember

The long a sound can be spalled at, a consonant e, or at. The long a round can be spelled ee, ea, e-consonant-e, y, or i-consonant-e.





Bieliow Level

Home/School Spelling Proctice Moster", TRB, p. 9





Above Lovel

Home/School Spelling Practice Moster", TRB, p. II







Answers For Unit 21 Spelling Connections

Janet P. Sitter, Ph.D.

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web ans the mole concept in chemistry is a fundamental concept that allows us to connect the mass of a substance to the number of particles atoms molecules or ions it contains it is based on avogadro s number which states that one mole of any substance contains $6\ 022\ x\ 10\ 23$ particles 2