Unit 08 - Stoichiometry - Worksheet 1

Answer the following questions in the space provided at the left.

18 mol O₂ 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?

2KCLO₃ → 2KCl + 3O₂

12 mol KClO₃ 3 mol O₂ 2 mol KClO₃

21 mol O₂ 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?

2KCLO₃ → 2KCl + 3O₂

14 mol KetO₃ 3 mol O₂ 2 mol KClO₃

__45 mol H₂O___ 3. How many moles of water are produced from 22.5 moles of oxygen?
2H₂ + O₁Z → 2H₂O

22.5 mol-O₂ 2 mol H₂O 1 mol-O₂

_20 mol HCI____ 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.

H₂ + Cl₂ → 2HCl

10 mol H₂ 2 mol HCl 1 mol H₂

21 mol Cl₂ 5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.

2Fe + 3Cl₂ → 2FeCl₃

14 mol EeCl₃ 3 mol Cl₂ 2 mol FeCl₃

Answer Sheet To Chemistry Quiz Review Stoichiometry

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Answer Sheet To Chemistry Quiz Review Stoichiometry:

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Unveiling the Magic of Words: A Report on "Answer Sheet To Chemistry Quiz Review Stoichiometry"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "Answer Sheet To Chemistry Quiz Review Stoichiometry," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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