

e. Pulmonary embolism

24. Drug metabolism

- a) Polar molecules cross mucous membrane easily
- b) Drug metabolism tends to make molecules more polar
- c) A base is less ionized in an acidic environment

This is a toxicology question in disguise! Has to do with enhanced elimination in toxicology.

Metabolism tends to make lipophilic drugs more polar (more excretable by the kidney) as ionized drug aren't reabsorbed.

Bases (ie most drugs), more ionized in acid environment so they are 'trapped' in acid urine and not reabsorbed. And acid drugs are more ionized in alkaline environment

Polar molecules tend not to cross mucous membranes

3. Formation of acetone occurs in poisoning with:

- a) ethanol
- b) methanol
- c) Ethyleneglycol ?
- d) isopropanol
- e) acetaminophen x

12. The following blood gases pH 7.17, PCO₂ 59, Bic 21, PO₂ 130 are most

likely to be consistent with:

- a) diabetic ketoacidosis
- b) diuretic overdose
- c) premature twin baby
- d) camphor ingestion with seizures
- e) oliguria and renal failure

15. One of the differences between morphine and pethidine relates to:

- a) analgesic efficacy
- b) route of metabolism
- c) ability to suppress cough
- d) histamine release effects
- e) abuse potential ?

68. All of the following substances bind well to activated charcoal EXCEPT:

- a) thioridazine
- b) atenolol
- c) cyanide
- d) benzotropine
- e) tetrahydrocannabinol (THC)

69. In paracetamol poisoning:

- a) there are theoretical advantages to giving the antidote at 2 hours post ingestion
- b) toxicity is less likely in children than adults
- c) the initial dose of N-acetylcysteine is given over 5 minutes
- d) N-acetylcysteine is not indicated more than 20 hours post ingestion
- e) activated charcoal is not used more than 1 hour after ingestion

Answered Mcq In Toxicology

Robin Proctor



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