#### **GCE Ordinary Level**

Paper 5070/01 Multiple Choice

| Question<br>Number | Key |
|--------------------|-----|
| 1                  | A   |
| 2                  | D   |
| 3                  | D   |
| 4                  | D   |
| 5                  | C   |
| 6                  | D   |
| 7                  | B   |
| 8                  | D   |
| 9                  | D   |
| 10                 | A   |
| 11                 | B   |
| 12                 | D   |
| 13                 | B   |
| 14                 | C   |
| 15                 | A   |
| 16                 | D   |
| 17                 | A   |
| 18                 | В   |
| 19                 | A   |
| 20                 | D   |

| Question<br>Number | Key |
|--------------------|-----|
| 21                 | 6   |
| 22                 | C   |
| 23                 | D   |
| 24                 | B   |
| 25                 | D   |
| 26                 | В   |
| 27                 | C   |
| 28                 | 100 |
| 29                 | C   |
| 30                 | C   |
| 31                 | C   |
| 32                 | C   |
| 33                 | A   |
| 34                 | D   |
| 35                 | B   |
| 36                 | A   |
| 37                 | - 8 |
| 38                 | B   |
| 39                 | D   |
| 40                 | C   |
|                    |     |

#### General comments

Almost all the questions discriminated well and only one question, Question 35, was found to be easy by all the candidates. The stem to each question is kept to a minimum and hence every word in the stem is of significance. Consequently it is very important that candidates read the questions carefully.

#### Comments on specific questions

#### Question 4

The most popular distractor was C. In sodium chloride the co-ordination number of the sodium ion is six, since it is not only the chloride ions in the same plane as the sodium ion but the chloride ions immediately above and below the sodium ion which are in contact with the sodium ion.



# **Answers For Chemistry Gce**

**Vijay Pandit** 

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Chapter 19 Moles and Equations Questions Chapter 20 Nitrogen and Sulfur Questions Chapter 21 Organic and Nitrogen Compounds Questions Chapter 22 Periodicity Questions Chapter 23 Polymerization Questions Chapter 24 Rates of Reaction Questions Chapter 25 Reaction Kinetics Questions Chapter 26 Redox Reactions and Electrolysis Questions Chapter 27 States of Matter Questions Chapter 28 Transition Elements Questions The Alcohols and Esters Quiz Questions PDF e Book Chapter 1 interview questions and answers on Introduction to alcohols and alcohols reactions The Atomic Structure and Theory Quiz Questions PDF e Book Chapter 2 interview questions and answers on Atom facts elements and atoms number of nucleons protons electrons and neutrons The Benzene Chemical Compound Quiz Questions PDF e Book Chapter 3 interview questions and answers on Introduction to benzene arenes reaction phenol and properties and reactions of phenol The Carbonyl Compounds Quiz Questions PDF e Book Chapter 4 interview questions and answers on Introduction to carbonyl compounds aldehydes and ketone testing nucleophilic addition with HCN preparation of aldehydes and ketone reduction of aldehydes and ketone The Carboxylic Acids and Acyl Compounds Quiz Questions PDF e Book Chapter 5 interview questions and answers on Acidity of carboxylic acids acyl chlorides ethanoic acid and reactions to form tri iodomethane The Chemical Bonding Quiz Questions PDF e Book Chapter 6 interview questions and answers on Chemical bonding types chemical bonding electron pair bond angle bond energy bond length bonding and physical properties bonding energy repulsion theory covalent bonding covalent bonds double covalent bonds triple covalent bonds electron pair repulsion and bond angles electron pair repulsion theory enthalpy change of vaporization intermolecular forces ionic bonding ionic bonds and covalent bonds ionic bonds metallic bonding metallic bonding and delocalized electrons number of electrons sigma bonds and pi bonds sigma bonds pi bonds s orbital and p orbital Van der Walls forces and contact points The Chemistry of Life Quiz Questions PDF e Book Chapter 7 interview questions and answers on Introduction to chemistry enzyme specifity enzymes reintroducing amino acids and proteins The Electrode Potential Quiz Questions PDF e Book Chapter 8 interview questions and answers on Electrode potential cells and batteries E Plimsoll values electrolysis process measuring standard electrode potential quantitative electrolysis redox and oxidation The Electrons in Atoms Quiz Questions PDF e Book Chapter 9 interview questions and answers on Electronic configurations electronic structure evidence ionization energy periodic table simple electronic structure sub shells and atomic orbitals The Enthalpy Change Quiz Questions PDF e Book Chapter 10 interview questions and answers on Standard enthalpy changes bond energies enthalpies Hess law introduction to energy changes measuring enthalpy changes The Equilibrium Quiz Questions PDF e Book Chapter 11 interview questions and answers on Equilibrium constant expression equilibrium position acid base equilibria chemical industry equilibria ethanoic acid gas reactions equilibria and reversible reactions The Group IV Quiz Questions PDF e Book Chapter 12 interview questions and answers on Introduction to group IV metallic character of group IV elements ceramic silicon oxide covalent bonds properties variation in group IV relative stability of oxidation states and tetra chlorides The Groups II and VII Quiz

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