

American Chemical Society (A.C.S.) General Chemistry Exam

75 multiple choice questions, 110 minutes

Bring a scanner, calculator and pencil(s)

- I. States of Matter**
Ideal Gas Law: $PV = nRT$, $PM = dRT$, deviations from ideal (high P, low T)
Ionic Solids: sc, fcc, bcc
Phase Diagrams: Pressure versus Temperature
Intermolecular Forces: ion-dipole, hydrogen-bonding, etc.
- II. Stoichiometry and Thermochemistry**
Hess' Law & Enthalpy: "standard states" have $\Delta H = 0$
Percent Composition: %C, %H, etc.
Limiting Reactant(s)
Stoichiometry
- III. Atomic Structure and Periodicity**
Oxidation States of "Main group" elements (Li, Be, B, etc.)
Oxidation Numbers
Atomic orbital configurations & relation to groups on periodic table (ns^2np^2 = halogens)
Periodic Trends: Electronegativity, atomic radii
Nuclear chemistry: mass number, atomic number
- IV. Molecular Structure**
Modes of bonding: ionic, covalent (both polar and nonpolar)
Lewis Dot, VSEPR, polarity, bond angles, resonance structures
- V. Solutions**
Molarity: mol solute / L solution, also calculating molarity, $C_1V_1 = C_2V_2$
Net ionic equations
Percent dissociation in water (electrolyte strength)
Molality: mol solute / kg solvent
Freezing Point Depression $\Delta T = k_{fp}$
- VI. Acids, Bases and Ionic Equilibria**
Know the "inert" acid/base ions: Cl^- , NO_3^- , Na^+ , etc.
Know general reactions for $WA + SB$, $WB + SA$, $SB + SA$
Equivalence point, finding unknown concentrations
Buffer: what makes a buffer, what a buffer is, etc.
pH: calculate pH of $WB + SA$, $SB + WA$, pure WB , pure WA
- VII. Chemical Equilibria - Molecular**
General equilibrium expressions: know how to find K
Precipitation reactions
Effect of heat upon endothermic / exothermic reactions
- VIII. Kinetics**
Rate of disappearance, rate of formation, Catalysis in mechanisms
Rate law and half-life from experimental data; Energy of activation
- IX. Thermo-dynamics**
Enthalpy, entropy, free energy
Relation of free energy to equilibrium constants
- X. Electrochemistry and Redox**
E values for electrochemical cells; E values for electrolytic cells
Balancing acidic redox reactions
Best reducing agents, oxidizing agents, definitions of oxidized and reduced
- XI. Descriptive Chemistry**
Nuclear chemistry: particle(s) emitted which result in increase of atomic number
Elemental reactivity: with water, oxygen, bases, acid, etc.
Organic chemistry: know alkenes, alkanes, etc.; also nomenclature
Coordination chemistry: coordination numbers around central metal
Ionization energy periodic trends
General nomenclature
- XII. Laboratory**
Effect of improper drying on mass measurements
Methods for finding $[Cl^-]$

American Chemical Society A C S General Chemistry Exam

I. Dwaine Eubanks



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and vocabulary that the American Chemical Society ACS Examinations Institute expects you to have mastered before sitting for your exam Test sections include Atoms Properties of Matter Bonding and Intermolecular Interactions Reactions Kinetics and Equilibrium Acids and Bases Thermodynamics Electrochemistry Nuclear Chemistry Safety Math and Data in the Laboratory and much more Our guide is full of specific and detailed information that will be key to passing your exam Concepts and principles aren't simply named or described in passing but are explained in detail The Mometrix general chemistry study guide is laid out in a logical and organized fashion so that one section naturally flows from the one preceding it Because it's written with an eye for both technical accuracy and accessibility you will not have to worry about getting lost in dense academic language Any test prep guide is only as good as its practice questions and answer explanations and that's another area where our guide stands out The Mometrix test prep team has provided plenty of general chemistry practice test questions to prepare you for what to expect on the actual exam Each answer is explained in depth in order to make the principles and reasoning behind it crystal clear Many concepts include links to online review videos where you can watch our instructors break down the topics so the material can be quickly grasped Examples are worked step by step so you see exactly what to do We've helped hundreds of thousands of people pass standardized tests and achieve their education and career goals We've done this by setting high standards for Mometrix Test Preparation guides and our ACS General Chemistry Study Guide ACS Exam Prep Secrets is no exception It's an excellent investment in your future Get the general chemistry review you need to be successful on your exam

The Hidden Curriculum - Faculty Made Tests in Science Sheila Tobias, 1997

[Diversity, Equity, and Inclusion for Mathematics and Science Education: Cases and Perspectives](#) Lin, Cheng-Yao, Sun, Li, 2025-06-04 Diversity equity and inclusion DEI are critical pillars for transforming mathematics and science education As classrooms diversify the need to address systemic barriers and create inclusive learning environments becomes more urgent Cases on DEI in STEM education highlight the real world challenges and strategies educators face in promoting equitable access to learning opportunities dismantling biases and empowering students from historically marginalized communities Further exploration may reveal powerful teaching tools and catalyze reflective practice and institutional change encouraging educators to critically examine their roles in shaping a more inclusive future in math and science Cases on Diversity Equity and Inclusion for Mathematics and Science Education explores key issues and concepts related to diversity equity and inclusion in mathematics and science classrooms It offers solutions and successful strategies for teaching and learning in mathematics and science This book covers topics such as inclusive classrooms K-12 education pre-service teaching and is a useful resource for educators sociologists academicians researchers and scientists

[Tests in Print](#) Oscar Krisen Buros, 1983

Building Synergy for High-Impact Educational Initiatives Janine Graziano, Lauren Chism Schmidt, 2016-05-17 Published in partnership with the Washington Center for Improving the Quality of Undergraduate Education First year seminars and learning communities are two of the most commonly offered high impact practices on U.S.

campuses The goals of these initiatives are similar helping students make connections to faculty and other students improving academic performance and increasing persistence and graduation As such it is not surprising that many institutions choose to embed first year seminars in learning communities This volume explores the merger of these two high impact practices In particular it offers insight into how institutions connect them and the impact of those combined structures on student learning and success In addition to chapters highlighting strategies for designing teaching in and assessing combined programs case studies offer practical insights into the structures of these programs in a variety of campus settings

Tests in Print III James V. Mitchell,1983 Preparing for Your ACS Examination in General Chemistry Lucy T. Eubanks,I. Dwaine Eubanks,1998 *Teaching Innovation in University Education: Case Studies and Main Practices* Saura, Jose Ramon,2022-06-17 In the last decade the development of new technologies has made innovation a fundamental pillar of education Teaching innovation includes the evolution of both teaching and learning models to drive improvements in educational methodologies Teaching innovation is a pioneer in the understanding and comprehension of the different teaching methodologies and models developed in the academic area Teaching innovation is a process that seeks validation in the academic and teaching communities at universities in order to promote the improvement and its practices and uses in the future characterized by digital development and data based methods Teaching Innovation in University Education Case Studies and Main Practices features the major practices and case studies of teaching innovation developed in recent years at universities It is a source on study cases focused on teaching innovation methodologies as well as on the identification of new technologies that will help the development of initiatives and practices focused on teaching innovation at higher education institutions Covering topics such as didactic strategics service learning and technology based gamification this premier reference source is an indispensable resource for pre service teachers lecturers students faculty administrators libraries entrepreneurs researchers and academicians

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Preparing for Your ACS Examination in Organic Chemistry I. Dwaine Eubanks,2002 Preparing for Your ACS Examination in General Chemistry - the Official Guide LUCY T. EUBANKS,2018-10-15 This guide is separated into first term and second term

general chemistry material Each section contains 8 chapters of material that also aligns to most general chemistry textbooks for a seamless addition to study materials for students Each chapter is designed with an introductory section of the material including common representations and where to find this material in a textbook The second section provides worked examples of typical multiple choice questions including how the correct answer is determined as well as how the incorrect answers were determined Also included for each study problem is a listing of the corresponding practice questions that use that concept The final section is a series of practice problems to test the concepts collectively The key is provided on a separate page for all study and practice problems

A Practitioner's Handbook for Institutional Effectiveness and Student Outcomes Assessment Implementation James Oliver Nichols, 1995

POGIL Shawn R. Simonson, 2023-07-03

Process Oriented Guided Inquiry Learning POGIL is a pedagogy that is based on research on how people learn and has been shown to lead to better student outcomes in many contexts and in a variety of academic disciplines Beyond facilitating students mastery of a discipline it promotes vital educational outcomes such as communication skills and critical thinking Its active international community of practitioners provides accessible educational development and support for anyone developing related courses Having started as a process developed by a group of chemistry professors focused on helping their students better grasp the concepts of general chemistry The POGIL Project has grown into a dynamic organization of committed instructors who help each other transform classrooms and improve student success develop curricular materials to assist this process conduct research expanding what is known about learning and teaching and provide professional development and collegiality from elementary teachers to college professors As a pedagogy it has been shown to be effective in a variety of content areas and at different educational levels This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor's role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get

started Appendices provide additional resources and information about The POGIL Project *Preparing for Your ACS Examination in Physical Chemistry* Thomas A. Holme, Kristen Murphy, 2009 Guide to Educational Credit by Examination Douglas R. Whitney, Andrew G. Malizio, 1987 **The Education Index** ,1990 *Process Oriented Guided Inquiry Learning (POGIL)* Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes Professional Careers Sourcebook Kathleen M. Savage, Charity Anne Dorgan, 1990 Provides a comprehensive overview of the literature and professional organizations that aid career planning and related research for 111 careers requiring college degrees or specialized education **The American Chemical Society at 125** ,2002

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Table of Contents American Chemical Society A C S General Chemistry Exam

1. Understanding the eBook American Chemical Society A C S General Chemistry Exam
 - The Rise of Digital Reading American Chemical Society A C S General Chemistry Exam
 - Advantages of eBooks Over Traditional Books
2. Identifying American Chemical Society A C S General Chemistry Exam
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an American Chemical Society A C S General Chemistry Exam
 - User-Friendly Interface
4. Exploring eBook Recommendations from American Chemical Society A C S General Chemistry Exam
 - Personalized Recommendations
 - American Chemical Society A C S General Chemistry Exam User Reviews and Ratings
 - American Chemical Society A C S General Chemistry Exam and Bestseller Lists
5. Accessing American Chemical Society A C S General Chemistry Exam Free and Paid eBooks

- American Chemical Society A C S General Chemistry Exam Public Domain eBooks
 - American Chemical Society A C S General Chemistry Exam eBook Subscription Services
 - American Chemical Society A C S General Chemistry Exam Budget-Friendly Options
6. Navigating American Chemical Society A C S General Chemistry Exam eBook Formats
- ePub, PDF, MOBI, and More
 - American Chemical Society A C S General Chemistry Exam Compatibility with Devices
 - American Chemical Society A C S General Chemistry Exam Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of American Chemical Society A C S General Chemistry Exam
 - Highlighting and Note-Taking American Chemical Society A C S General Chemistry Exam
 - Interactive Elements American Chemical Society A C S General Chemistry Exam
8. Staying Engaged with American Chemical Society A C S General Chemistry Exam
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers American Chemical Society A C S General Chemistry Exam
9. Balancing eBooks and Physical Books American Chemical Society A C S General Chemistry Exam
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection American Chemical Society A C S General Chemistry Exam
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine American Chemical Society A C S General Chemistry Exam
- Setting Reading Goals American Chemical Society A C S General Chemistry Exam
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of American Chemical Society A C S General Chemistry Exam
- Fact-Checking eBook Content of American Chemical Society A C S General Chemistry Exam
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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