Particulate Nature of Matter

<u>Atom</u> – the smallest unit of matter and having a single neutral charge.

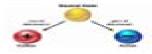


Molecule – the neutral particles made of two or more atoms bonded together. The smallest fundamental unit of a chemical compound that can take part in a chemical reaction.



<u>lon</u> – An atom or group of atoms that carries a positive or negative electric charge as a result of having lost or gained one or more electrons.

- <u>cation</u> positively charged ion: loses electron.
- <u>Anion</u> negatively charged ion: gains electron.



States of Matter

Solid – Has a definite shape and volume.



<u>Uguid</u> – Has a definite volume but takes the shape of its container.



Gas – Has no definite shape or volume.



Plasma - It is an ionized gas.



Bose-Einstein condensate – A boson particle is cooled to near absolute zero and has very low energy.

Fermionic condensate - A fermion particle is cooled to near absolute zero and has very low energy.

<u>Superconductor</u> – Conduct electricity with virtually zero electrical resistance: Once a current is started, it flows indefinitely.

Superfluid - The liquid flows resisting friction.



Properties of Matter

<u>Physical properties</u> – These are qualities of matter that are observable and measurable without changing composition.

- Intensive properties Properties that do not depend on the amount of matter present.
 - Color
 - Odor
 - Luster
 - 4. Malleability
 - Conductivity
 - Hordness
 - Melting/freezing point
 - 8. Boiling point
 - 9. Density
- Extensive properties Properties that depend on the amount of matter present.
 - Moss.
 - Weight
 - Volume
 - 4. Length

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D.Antonio Cantu, Wilson J. Warren

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